

Euro organicchemistry 2019: Title: synthesis of new structures of imidazolium salts

Eyad Mallah

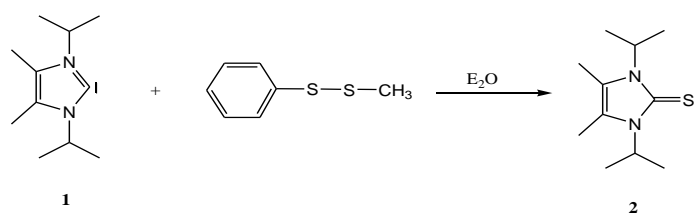
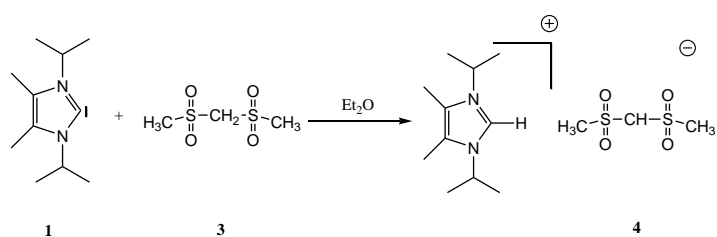
Faculty of Pharmacy, University of Petra, Amman- Jordan

Abstract

As result of the strongly basic character of heterocyclic carbenes **1**, they react with *Broenstedt* acids and have consequently been used as selective deprotonation reagents. The 2H-imidazolium salts formed by this method are accessible by other routes, alkylation of 2H-imidazoles, cyclization reactions or from the thiones and nitric acid, and may be used as precursors in the synthesis of **1** through deprotonation.

So our current research efforts also continue to focus on design and synthesis of new structures of imidazolium salts. Therefore, Owing the strongly basic character of heterocyclic carbenes reacted with methyl phenyl disulfide to give the corresponding adduct **2** which afford a new synthetic route for 2,3-dihydro-imidazole-2-thione. While the reaction of **1** with bis-methane sulfone was carried out at RT to give the salt **4**.

The above mentioned reactions are outlined in scheme 1, the results confirmed by NMR, mass spectroscopy, elemental analysis and single crystal X-ray diffraction.



Synthesis of precursors of coupling reactions : Preparation of 1-phenyl-1H-imidazole

Cold acidic corrosive (30.25 mL, 528.47 mmol, 4.3 eq), watery formaldehyde (9.04 mL, 328.14 mmol, 2.67 eq), and watery glyoxal (13.87 mL, 302.3 mmol, 2.46 eq) were moved to a round base flagon (150 mL) and warmed at 70 °C. A solution of cold acidic corrosive (30.25 mL, 528.47

mmol, 4.3 eq), ammonium acetate in water (9.47 g/ 6.15 mL), and aniline (11.2 mL, 122.9 mmol, 1eq) was added drop-wise to the flagon over a time of 1 hour. The arrangement was constantly mixed and warmed at 70 °C for 18 h. The response blend was at that point cooled to room temperature and added dropwise to a stirred arrangement of NaHCO₃ (88.9 g) in water (900 mL), and the watery layer was extricated with diethyl ether (3x150 mL). The consolidated natural

layers were washed with salt water, dried over anhydrous MgSO₄ and vanished in vacuo. The joined natural concentrates were concentrated and the subsequent buildup was sanitized by section

chromatography on silica gel (Petroleum ether/ EtOAc = 20:80) to yield 1-phenyl-1H-imidazole as

a yellow oil (7.48 g, 42 % yield).

HRMS (ESI) Calcd for C₉H₈N₂ [M+H]⁺ : 145.0760. Found : 145.0760

C₉ H₈ N₂ (144.17). R_f (AcOEt, 100 %) = 0.4 Flash chromatography eluent PE/AcOEt, 20/80. Yellow oil. Yield = 42 %

Antimicrobial activity

All incorporated imidazolium salts were tried for antimicrobial action against microscopic organisms and yeast. Least inhibitory focus (MIC) estimations of the ILs and reference antimicrobial specialists were appeared. It

was resolved that the DMSO had no movement against test microorganisms.

The imidazolium salts subbed with alkyl chains of a shorter length had poor surfactant properties, and resultant poor MIC values true to form. Just IL2 indicated great antifungal action as other ionic fluids containing long alkyl chains. It is seen that the MIC estimations of the imidazolium salt bearing ethoxy ether gathering (IL3), was no change against bacteriae and parasite (>14.6 mM). Then again, MIC estimations of IL2 were 1.8-15mM. The antimicrobial exercises of IL3 were bad when thought about IL2. This circumstance can be clarified by the length of the alkyl chain; the two mixes have same chain length. This has been appeared in past investigation [Cn-Im-3OEG][Cl] mixes demonstrated better exercises with longer alkyl chain against bacteriae. Other than it is demonstrated that antimicrobial exercises against bacteriae expanded relying upon the length of alkyl chain in imidazolium salts.

Then again, the two mixes showed increment on poisonousness towards the Gram positive sort microorganisms (*S. aureus*) contrasted with Gram negative. These outcomes might be deciphered as the distinctive basic properties present in the two kinds of bacteriae. The external layer in bacterial cell dividers is available in Gram positive, while the external film isn't existent in Gram negative. A few segments (lipopolysaccharide later engraving) of Gram negative dividers can monitor the bacteriae from synthetic substances.

Imidazolium salts fill in as the atomic skeleton in numerous mixes with anticancer action and some of them indicated a repressed impact of PI3K. A progression of imidazolium salt subordinates were planned and orchestrated by sub-atomic hybridization devices in the earlier research, with the cross breed compound showing strong cytotoxic movement against HL-60, A549 and MCF-7 tumor cell lines (the 77 crossover mixes with the mean IC50 estimations of 2.84 μ M) There was no further structure-work relationship, target or system as for these novel imidazolium salt subordinates.

Basic adjustment of a recognizable characteristic item, dynamic compound or clinical medication is a proficient technique for structuring a novel medication. The fundamental reason for basic adjustment is to decrease the harmfulness of target compound, while improving the utility of the medication [20]. This is commonly done by adjusting the key substituent bunch in the atomic skeleton of target mixes to build the coupling fondness and explicitness to the dynamic site of receptor protein, and improve ADME (retention, appropriation, metabolic and discharge), and changing the lipid-fluid parcel. The most significant advance in sedate plan is to foresee the objective of a given compound and research the coupling proclivity for and explicitness to the dynamic objective, which is feasible through the use of Computer-Aided Drug Design (CADD) procedures, which can improve the proficiency of this procedure

Target distinguishing proof is an essential advance in the medication plan pipeline and procedure, and utilizes PharmMapper. PharmMapper is a uninhibitedly open electronic instrument that is used for foreseeing the potential medication targets through a "turn around pharmacophore" (otherwise called "target angling") mapping technique. Profiting by an exceptionally proficient and hearty mapping technique, PharmMapper, with its high-throughput capacity, can distinguish the potential objective up-and-comers from the database with a runtime of a couple of hours.

Biography

Dr. Eyad Mallah has completed his PhD at the age of 30 years from Tuebingen University, Germany. I'm currently working as an associate professor and Dean's Assistant at Faculty of Pharmacy in University of Petra.

I have a specific interest in Pharmaceutical organic chemistry and Pharmaceutical analytical chemistry. My degree has provided me with a strong background in all areas of Pharmaceutical Chemistry and I have developed a specific interest in bioequivalence studies since I have worked in this field at (JCPR) bioequivalence center for about 7 years. I'm currently working on development of chromatographic and immunoassay methods for analyzing different pharmaceutical compounds in biological fluids which could be used in bio-studies. In

addition, I'm interested in drug-drug interactions and drug-juice interactions which could affect the pharmacokinetic profile of some pharmaceutical products. I'm also working on the synthesis of some Barbituric acid and Imidazole derivatives and their application in pharmaceutical compounds. I have already published about **50** articles in the above disciplines in different international Journals. (Up to 100 words)