

Nano Congress 2019: Study of phase transformation of Cobalt nanowires at high cobalt ion concentration prepared through template based deposition - Aiman Mukhtar - Wuhan University of science and technology

Aiman Mukhtar

Wuhan University of science and technology, China

Among different synthesis techniques electrochemical synthesis of nanowires in anodic aluminium oxide (AAO) template allow control over composition and size of NWs. AAO template are made by anodizing the aluminum foils in oxalic acid (COOH)₂. In this study the phase transformation of Cobalt nanowires was observed at high cobalt ion concentration. The cobalt nanowires was electrochemically deposited in AAO template with direct current (-1.6V) deposition technique using three electrode bath cell. The average pore diameter of nanowires was 50nm which was equal to nanopores in AAO template. XRD results show that using 0.356M solution the hcp Cobalt nanowires were formed at high concentration transformation was occurred. The Fcc Cobalt nanowires were formed at 1.067M solution. The transient (current Vs time) curves display with increasing concentration of Co²⁺ ions, the *i*_{max} increases while the *t*_m decreases. Shorter *t*_m and a higher *i*_{max} value can represent a larger *N*_s (saturation nucleus density). Therefore we believe that the electrochemical deposition at high concentrations of 1.067M at -1.6V could lead to large *N*_s, representing the formation of smaller critical nuclei. The structure of Co can be determined by the critical nucleus size and smaller critical nuclei favour the formation of fcc Co. Therefore the fcc Co nanowires were observed when depositing in the high cobalt concentration solution.

Introduction

Co and Ni composite stage chart shows that [1], Co and Ni display total strong arrangement in the fcc stage at temperatures between the solidus and the allotropic change temperature, compound stage outline affirms that change happens at the temperature around 300 °C. Along these lines, it is realized that fcc Co_{84.43}Ni_{15.51} is a high-temperature stage. Our past research articles show that structure of Co relies upon affidavit parameters. High potential and electrolyte focus favors fcc Co stage while low potential and electrolyte fixation favors hcp Co stage. The ongoing article proposed another model which portrays the arrangement of fcc Co with DC testimony at the high capability of - 3.0V.

A great deal of research work has been done on CoNi amalgam nanowires with respect to their concoction arrangement and structure in AAO layout.

Numerous scientists found the blended period of hcp and fcc at 30:70 nickel cobalt particle focus in an electrolytic arrangement utilizing AC Electrodeposition. Nonetheless, no work has been done on the impact of potential on stage change of CoNi amalgam nanowires. The flow explores endeavors to examine the impact of potential on the precious stone structure and synthesis of Co-Ni composite nanowires. Expectation that this examination would be useful to create other ferromagnetic compound nanowires utilizing AAO layout in future.

Conclusion

The current examination shows that the AAO layout with intermittent hexagonally masterminded nano-pores with in all around adjusted CoNi nanowires utilizing various possibilities is effectively manufactured. Results show that the Phase change of CoNi compound nanowires happens at the high capability of - 3.5 V. The high potential fcc stage is a metastable stage, while the CoNi composite nanowires saved at low potential has hcp stage. The arrangement of fcc combination nanowires can be credited to littler basic groups shaped at the high potential.