Current environmental mercury poisoning of children: A literature review of its impact on global pediatric health.

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Abstract

Mercury is a ubiquitous environmental pollutant with diverse adverse health effects that could result in death. Children are the most susceptible population of the world to mercury poisoning. Two earlier reviews by the author dealt with fatal mercury poisoning cases in the general population and mercury vapor exposure in dentistry. The objective of the present article is to review global studies on environmental mercury poisoning of children and get perspective on its impact on pediatric health by assessing the progress made in diagnosis, treatment, prevention and monitoring of the adverse effects. With the aid of several search platforms, studies of environmental mercury exposures in children published in the world literature were collected. With emphasis on clinical studies, details of the reported adverse health effects and treatments used were compiled along with prevention and monitoring aspects of mercury exposures. Mercury poisoning events were categorized based on exposure sites (at home or outside), its form (elemental, organic or inorganic), route (inhalation, oral ingestion, skin absorption or pre-and postnatal), duration (acute or chronic) and dose. The clinical signs, symptoms and treatments used in each category were separately enumerated for comparative purpose. Home is the most common site for children's exposure to mercury attributable to breakage of fever thermometers, dental amalgam fillings, tainted cosmetics, toys and jewelry, and consumption of contaminated fish, OTC and herbal medicines, and dietary supplements. The main organs affected are brain, lungs, kidneys and immune systems. Clinical interpretation of blood and urine levels of mercury are unambiguous when they are high and become difficult as they approach normal range. While diagnosing mercury poisoning can be challenging, it can be made with reasonable reliability and promptly treated with chelation therapy. With the development of mercury-free products and manufacturing processes along with industrial pollution abatement measures, children's exposure to mercury is currently being reduced. Parents, pediatricians, and school science teachers can play a major role in preventing mercury poisoning of children. This review should be of immediate interest to environmental scientists and regulators around the world.

Keywords: Mercury, Environment, Children's Health, Diagnosis, Prevention, Monitoring.

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Abbreviations: AA: Arachidonic Acid: AAP: American Academy Of Pediatrics; ADHD: Attention Deficit Hyperactivity Disorder; AHRQ: Agency For Healthcare Research And Quality (US); ASD: Autism Spectrum Disorders; ASGM: Artisanal and Small-Scale Gold Mining; ATSDR: Agency for Toxic Substances and Disease Registry (US); BAEP: Brainstem Auditory Evoked Potential; BASC: Behavior Assessment System for Children; BBB, Blood-Brain Barrier; BSID: Baby Scales of Infant Development; CDC: Center for Disease Control and Prevention (US); CHD: Coronary Heart Disease; CNS: Central Nervous System; CPT: Continuous Performance Test; DHA: Docosahexaenoic Acid; DMPS: 2,3-Dimercapto-1-ropanesulfonate; DMSA: Dimercaptosuccinic Acid; DNA: Deoxyribonucleic Acid; EFSA: European Food Safety Authority; EPA: Environmental Protection Agency (US); ER: Emergency Room; EU: European Union; FAO: Food And Agricultural Organization (WHO); FDA: Food And Drug Administration (US); FLB: Fluorescent Light Bulb; FTT, Finger-Tapping Test; GDS, Gesell Developmental Schedules; GI, Gastrointestinal; GMDS, Griffiths Mental Development Scales; HVAC: Heating, Ventilation and Air Conditioning; LED: Light Emitting Diode; MDI: Mental Developmental Index; MeHg: Methylmercury; MRI: Magnetic Resonance Imaging; MSCA, McCarthy Scales Of Children's Abilities; NHNE, National Health And Nutrition Examination (US); NIEHS, National Institute Of Environmental Health And Safety (US); NIH: National Institutes of Health (US); NIOSH: National Institute Of Occupational Safety And Health (US); NSA: National Science Foundation (US); OTC: Over-The-Counter; PDI: Psychomotor Developmental Index; PHS: Public Health Service US); PPVT: Peabody Picture Vocabulary Test; PUFAs; Polyunsaturated Saturated Fatty Acids; RFD: Reference Dose; RI: Risk Index; ROS: Reactive Oxygen Species; SACMEQ: Southern and Eastern African Consortium for Monitoring Educational Equality; SBIS: Stanford-Binet Intelligence Scale; Se: Selenium; SH: Sulfhydryl; TGMD: Test for Gross Motor Development; TMT: Trail Making Test; TWI: Tolerable Weekly Intake; UK: United Kingdom; UNFAO: United Nations Food and Agriculture Organization; US: United States (of America); VEP: Visual Evoked Potential; VER: Visual Evoked Response; VRM: Visual Recognition Memory; WHO: World Health Organization; WISC: Wechsler Intelligence Scale for Children and WRAVMA: Wide-Range Assessment of Visual Motor Abilities.

Introduction

Children are the most susceptible population of the world to environmental mercury poisoning. The two major routes of mercury exposure are inhalation of its vapor in the polluted air and consumption of fish and seafood contaminated with its organic form, MeHg. According to WHO [1], the ubiquitous and persistent nature of mercury and its compounds pose a threat for the healthy development of the world's children. The Minamata Convention [2] and EU Science for Environmental Policy [3] are currently addressing pediatric health problems raised by the complicated life cycle of mercury in the global environment. Study of early-life exposure to mercury and later-life diseases is a priority of the developmental origins of health and disease program of the US NIEHS [4] and WHO [5]. Also, WHO and the UN Environmental Programme have initiated global monitoring of children's exposure to mercury [6]. In recent years, notable progress has been made in curtailing exposure of children to this important environmental pollutant and monitoring the health of those exposed to it [7].

Mercury poisoning in the general population that resulted in death [8] and its vapor exposure in dental practice [9] were the subject of two earlier reviews by the author. The present review deals with pediatric mercury poisoning of children ≤ 18 yr. of age.

Historical perspective

Human exposure to mercury is due to its presence in the global environment from natural sources (69%) and continued human contributions (31%) from its industrial, commercial, medical and dental usages over the past 200+yr [10-19]. The earth's crust contains 0.05 mg mercury/kg and the current estimate of global mercury emissions is 7,527 Mg/yr. [20]. In the bioenvironment, mercury is transformed by microorganisms (plankton) to MeHg [21] and bioaccumulated in small fish feeding on it [22]. MeHg is further biomagnified in large predatory fish and other marine mammals ingesting these small fish [23]. Human contribution to the present-day mercury in Arctic marine animals used as food is estimated to be 92.4% [24]. The Arctic populations have blood mercury levels among the highest in the world that are associated with adverse health outcomes across life stages (from neurodevelopment in infancy to CHD in adults) [25]. The climate-induced amplification of MeHg in predator fish is projected to reach 8% by 2100 [26].

Mercury-containing industrial discharges (in addition to the atmosphere), agricultural runoffs (containing mercury-based fertilizers and pesticides) and domestic sewage (with mercury from cosmetics, OTC drugs, etc.) are the major polluters of lake, river and other large bodies of water [27-29]. Socioeconomic and health consequences of mercury pollution, and medical benefits of reducing mercury exposure in humans are current issues of global interest [30-33].

Mercury is devoid of any known physiological benefit and its exposure in any form (elemental, organic or inorganic) is considered potentially toxic to humans [34,35]. However, a physiological role for Hg++ during phototropic growth of the purple non-sulfur bacteria has been recently reported [36]. Based on an estimated ambient air level of 10 ng/m³ mercury, its average daily intake in humans by inhalation is ~0.2 μ g and from drinking water containing 0.5 μ g mercury/L it is ~1 μ g [37]. However, food is by far the largest contributor to human mercury exposure (2-20 μ g/day) with fish and seafood being the major sources.

In the current global pandemic of corona virus disease 2019 caused by SARS-CoV-2 (COVID-19), reduction of mercury exposure may be considered as a potential tool for lowering vulnerability and severity of this deadly respiratory viral infection [38-41]. An unusual and paradoxical feature of the current COVID-19 pandemic is that children appear to be less severely affected by this virus than adults [42].

Children's exposure to mercury

The common forms and sources of mercury relevant to pediatric exposure are:

- 1. Elemental mercury, usually as vapor from mercury spills [43,44], dental fillings [45], religious practices [46] and tainted toys and jewelry [11],
- 2. Organic mercury compounds, MeHg being the most common through consumption of contaminated fish and seafood [47-49],
- 3. Maternal use of inorganic mercury compounds, such as bromide and chloride salts in cosmetics, household products, herbal medicines and dietary supplements [50,51].

The form of mercury a child is exposed to has significant effect on its toxic manifestation since there are major differences in the body uptake, transport and disposition of the three common forms of the metal [8,34,52]. Often children are simultaneously exposed to different forms of mercury which are generally chronic in nature and adverse effects tend to be definitive at higher doses but subjective at trace levels [1,18].

In children, vital organs (such as lung, brain and kidney, and immune systems) are at critical stages of development and often targets of adverse effects of mercury exposure [22,53-55]. This could occur even before birth since mercury readily passes through the placenta and enters the developing fetus during gestation in exposed pregnant women [56,57].

AAP has provided information on children's mercury exposure for practicing pediatricians [58] and the US EPA [59] has published guidelines to physicians in conducting mercury medical surveillance programs. Also, WHO guidance for identifying populations at risk from mercury exposure is available [60].

Alarmed by the reports of residual mercury vapor exposure of children in daycare centers and new condominiums which were converted from industrial buildings which used mercury and inadequately remediated, the US Congress in 2008 directed ATSDR and CDC to form the Mercury Workgroup to investigate such mercury exposure in children. In 2009, the Workgroup published a comprehensive report with the following conclusions [61]: Children as a group are more sensitive to mercury vapor exposures in contaminated spaces and at higher risk than adult population. Since they breathe at a faster pace, have larger lung surface area relative to body mass than adults, and their shorter stature, crawling and play activities keep children's breathing zone closer to contaminated floor. The duration of such exposure (acute or chronic) and concentration of mercury govern the severity of adverse effects.

While environmental mercury poisoning of children around the globe continues to be reported to-date, there appears to be no literature review of the topic over a decade.

Literature Review

According to the 2020 annual report by the American Association of Poison Control Centers in the US, 1,965 calls pertaining to mercury were received with 1,256 involving broken fever thermometers [62]. Accidental mercury poisoning of children continues to be of public health concern in the US and around the world [63-71]. The objective of this review is to assess the current status of global mercury poisoning of children, provide up-to-date information on how to diagnose and treat, and appraise preventive and monitoring methods. This should be of interest to environmental scientists and regulators as well as pediatricians, medical personnel in hospital emergency rooms and poison control centers. Also, the information gathered should be useful to first responders and school science teachers in educating parents and their children how to minimize mercury exposure.

A literature search on the topic of mercury poisoning (key words: environment, mercury, poisoning and children) was performed using PubMed, MEDLINE, Academia-Edu, Science Direct and Google Scholar search platforms. Clinical studies published through June, 2023 were the main focus of this review. Those deemed relevant were used to collect detailed data on where, how, and the form (elemental, organic or inorganic) and nature (acute or chronic) of mercury exposures reported in children. Also, children's ages, biological samples used as biomarkers, clinical signs and symptoms of adverse effects, treatments and autopsy results (in the case of fatalities) were gathered. Available historical and chronological data on mercury poisoning were also compiled for possible extrapolation and its relevance to current events.

Results

Over 50,000 articles on the general topic of toxic effects of mercury exposures in children were available online. Of these, 534 were identified as relevant to this review objective and details of the reported exposure events were obtained. In general, the robust studies from Brazil, Canada, EU (primarily Scandinavia and Spain), UK and US were a salient part of this review. They were separated into various exposure categories based on different forms of mercury, common sites and sources of contaminations. Details of mercury concentrations in various biological and tissue samples used as biomarkers were collected. The reported clinical signs, symptoms and treatment of mercury poisoning was gathered under elemental, organic and inorganic forms of mercury.

In the US, EPA has recognized the following neurodevelopmental disorders in children attributable to environmental pollutants [72]: vision and hearing impairments, ADHD, learning and intellectual disabilities, ASD and cerebral palsy. Also, WHO recognizes the above disorders in evaluating health risk of environmental pollutants in exposed children [73].

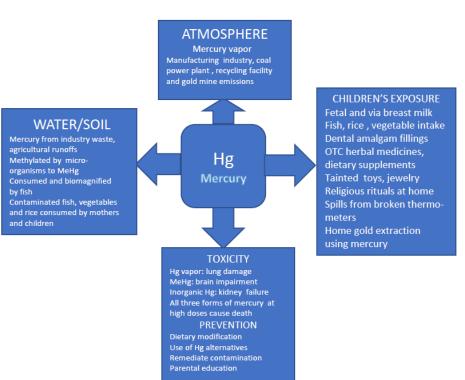
From conception through adolescence, rapid growth and developmental processes occur in children which can be disrupted by exposure to mercury in the environment [1,22,53]. This occurs through: 1) maternal exposure, prenatally in the uterus and postnatally *via* breast milk, 2) inhalation, 3) oral ingestion and 4) skin contact.

Mercury affects the reproductive function of men as well as women [74]. During gestation, both elemental mercury and MeHg readily cross the placental barrier and accumulate in the fetus [75,76]. Mercury body burden of the mother is shared by her fetus and neonate which may result in larger exposure doses due to their lower body weights. Also, children consume more contaminated food and beverages for their body weight than adults. Overall, compared to adults, children have more years of future life and thus more time to develop chronic diseases that may take decades to manifest (lag time) after mercury exposure [76-80].

Inhaled elemental mercury vapor is readily absorbed in lungs (~80%) and crosses the BBB. In the brain, it is oxidized to inorganic mercury (Hg++) and bound to macromolecules (DNA, enzymes, proteins, etc.) *via* SH groups. However, elemental mercury is slowly absorbed through skin and little is absorbed in the GI tract when ingested orally. In contrast, organic MeHg is readily absorbed through skin contact as well as in the GI tract (90%-95%) when ingested and crosses the BBB, biotransformed to Hg++ by demethylation and bound to macromolecules in the brain [81-84]. On a cellular level, MeHg induces oxidative stress by elevating ROS [85]. Since inorganic mercury compounds are poorly absorbed by dermal and oral routes (~10%), and do not cross the BBB, the main site for their accumulation is the kidney [55].

Inhaled elemental mercury vapor is excreted from the body in the exhaled breath, sweat and urine. Ingested mercury in liquid form is excreted in the feces unchanged [86,87]. Also, oral MeHg exposure leads to its excretion in the feces as inorganic mercury, since the gut flora can demethylate MeHg and modify its absorption and bioavailability [82]. The half-lives of the three common forms of mercury for elimination from the body are: elemental mercury (58 d), inorganic mercury (30-60 d) and MeHg (70-80 d) [83].

Elemental mercury generally does not accumulate in food [88] and MeHg usually not present in drinking water [36,89]. Diet and nutritional status of children (see Figure 1) have major impact on mercury toxicity [1,86,90,91].



GRAPHICAL ABSTRACT

Figure 1. Graphical abstract highlighting the origin of children's exposure to elemental mercury, its organic and inorganic compounds and modulatory effect of diet and nutrition on mercury toxicity.

Discussion

The following are common sites for exposure of children to elemental mercury, and its organic and inorganic compounds:

Mercury exposure at homes

Home is the most common site for children's exposure to mercury. Such exposure occurs due to: 1) consumption of mercury contaminated fish and other dietary constituents, 2) breakage of mercury-containing devices, *e.g.*, fever thermometer, 3) heating mercury during unauthorized smelting operation by parents to extract gold and silver, 4) cultural or religious ceremonial utilization of mercury, 5) household use of unregulated cosmetics, herbal medicines and dietary supplements, and 6) family members unknowingly tracking mercury home from workplace. Additionally, children bringing home mercury from school science laboratories and abandoned factories could lead to the toxic metal vapor exposures (See below Sections From acute exposure to high doses of mercury and Mercury exposure at other locations, respectively).

From consumption of MeHg contaminated fish by mothers

Pregnant mothers who consume contaminated fish expose their offsprings to mercury during gestation and postnatally, through breastfeeding [48,49,92-94]. Mothers with lower body weight (<60 kg or ~132 lb) generally have higher mercury levels in their breast milk [95]. Extended breastfeeding (up to 3 yr.) is not significantly associated with elevated mercury level in children as measured in their scalp hair [96]. Interestingly, human breast milk contains mercury 8x WHO guideline amount for drinking water (1 μ g/L), 4x that allowed in bottled water (2 μ g/L), and

higher than in dairy cow milk (<2 μ g/L) [97] and infant formula milk powder (<0.5 μ g/g) [98]. However, there is consensus that the benefits of breastmilk for infant outweigh any risk of mercury exposure [95,99-101].

The lactational exposure of infants to MeHg, a developmental neurotoxin [83,102,103], is reduced to \sim 50% at 2-3 mths of age compared to that expected on the basis of maternal blood levels [49,104]. During this early growth period, infant body weight gains rapidly (\sim 1.5-2x) and the resulting increase in body volume appears to dilute MeHg levels (and lower toxicity) in breastfed infants.

While fish provide healthful dietary proteins and nutrients, such as omega-3 long-chain PUFAs (e.g., DHA and AA) [92,105,106], these benefits may be offset by MeHg contaminant [107-109]. Fish contain 90%-95% of its total mercury as MeHg [110-112]. The EPA RfD for chronic MeHg exposure from diet is 0.7 µg/ kg bw/wk (=mercury in blood, 5.8 μ g/L and scalp hair, 1.0 μ g/g) for optimal health [113]. The provisional TWI of MeHg set by various regulatory agencies is 0.7-1.6 µg/kg bw [108,114]. However, in Japan where fish consumption is highest in the world, TWI for MeHg is 2.0 µg/kg bw [115] and its weekly average lactational exposure in infants is 0.63 µg/kg bw [116]. In 17 EU counties, mothers who consumed at least one fish meal/wk had hair mercury level $\leq 0.55 \ \mu g/g$ with about half of that in their children's hair [113]. In contrast, the Taiwanese infants have elevated mercury level (79 ng/g stool) at birth due to high maternal intake of mercury (>100 µg/mo.) from fish diet during pregnancy [117,118].

DHA and child development

DHA is considered a critical nutrient in the development of

nervous system and infant growth in the early life during gestation and while breastfed [76,105,119-126]. Both FDA and EPA recommend nursing, pregnant and childbearing-age women to consume twice a week seafood containing low MeHg and high DHA levels (*e.g.*, shrimp, pink salmon, seabass, mackerel, whiting, sole, pomfret and trout), and avoid those with low DHA and high MeHg (*e.g.*, whales, swordfish, shark, tuna, halibut, walleye, burbot and pike) [127-130]. Also, AAP, American Heart Association and American College of Cardiology recommend 200-300 mg DHA/day or 1-2 servings of fish/wk [131,132]. A Belgian study found such fish consumption to be safe for increasing the maternal DHA intake [133].

Worldwide human breast milk analysis found a wide range for DHA content (0.06%-1.40%) by wt.; optimal, 0.2-0.5%) with highest concentrations in seafood-consuming coastal populations [134]. Thus, breastfed children of mothers in Menorca, Spain with fish intake >2-3 times/wk during pregnancy (particularly during the third semester spurt of fetal brain development) and lactation achieved significantly higher MSCA scores when tested at age 4 yr. [135]. Additional studies also support such beneficial effects of breast milk DHA in children attributable to maternal fish consumption [136-138]. In the US, only ~25% of pregnant women consume the recommended quantity of fish needed to achieve DHA intake for optimal child health [139].

Since breastmilk DHA levels decrease during 12 wks postpartum, its supplementation is suggested for nursing mothers [140,141]. Thus, DHA supplement *via* maternal cod fish intake (twice a wk) has enhanced infant development [142-145]. Also, fish oil supplement has enhanced breastmilk DHA levels in nursing Australian women and development of their children (based on Griffiths score) [146]. Further, the long-range (at 2-4 yr.) developmental benefits of maternal DHA and fish oil supplements have been confirmed by an analysis of 34 clinical trials involving >16,000 breastfed infants [147]. Such supplementation is endorsed in EU [148] and Asia [149]. Nevertheless, studies in India [150], Italy [151] and Netherlands [152] have reported no such benefits of maternal DHA supplementation or fish consumption on child development in the short range (at 1-1.5 yr.).

Prenatal MeHg exposure

While moderate fish intake (1-3 times/wk; recommended by FDA and EPA) during pregnancy improves the metabolic health and inflammatory biomarkers in children, such intake >3 times/wk could lead to unfavorable metabolic profiles [153]. A study of the 14-yr.-old children in the Faroe Islands found an association between CPT (a measure of the speed of visual information processing) and high prenatal MeHg exposure from fish consumption during pregnancy (maternal hair mercury >10 μ g/g) [154]. This was attributed to MeHg-induced dysfunction in the brain frontal lobes. Also, the survey of a large group of 44,824 Danish pregnant women with high consumption (60 g/day) of fatty fish (*e.g.*, salmon, trout and halibut) reported reduced fetal growth [155]. However, this study down played the role of MeHg and ascribed the results to other persistent organic pollutants in fish.

The Inuit children of high-fish consuming women in Arctic Quebec, Canada with cord blood mercury >7.5 $\mu g/L$ were found

to be 4x as likely to have IQ<80, the clinical cut-off score for borderline intellectual disability [156].

Interestingly, an analysis of MeHg in 130 placentas at birth in Kuopio University Hospital, Finland found higher amounts in primiparas (first time mothers) which increased with maternal age [76]. A study in Krakow, Poland found that the prenatal exposure to mercury (mean blood levels, maternal >0.5 μ g/L and umbilical cord, 1.05 μ g/L) from fish consumption, especially during the last trimester of pregnancy, delayed cognitive and psychomotor functions (using BSID scale) in 1-yr-old infants [157,158]. Other factors such as mother's age, country of origin, smoking and season of delivery were also significantly and independently associated with cord blood mercury concentrations [159]. An additional study found that smoking accelerated the loss of brain neurotrophic factor in newborn induced by maternal MeHg intake from whale diet [160].

In Mexico City, mercury levels in pregnant women (3.4 μ g/L blood and 0.5 μ g/g scalp hair) and children (1.8 μ g/L blood and 0.6 μ g/g scalp hair) have been reported [161] These mercury levels are 3-5x higher than those in the US and Canada primarily due to the consumption of seafood contaminated with mercury >EPA guidance value, 0.3 μ g/g.

While prenatal mercury exposure is associated with greater risk of ADHD-related behaviors in children, fish consumption during pregnancy appears to provide protection against such behaviors [162-169]. There is ongoing discussion on the adverse effects of maternal mercury exposure on fertility, reproductive health and pregnancy outcomes [74,170]. Based on current evidence, dietary mercury exposure during pregnancy (blood range, 0.64-3.71 µg/L and hair levels, 0.3-5.7 µg/g) is unlikely to be a risk factor for neurodevelopmental deficits in early childhood (0-5 yr.) [171]. In this context, the potential study bias [1672] and difficulties in drawing definitive conclusions using neurophysiological tests in children exposed in utero to low levels of MeHg [173] are worth noting. (See Effect of dietary constituents... section below).

Genetics of prenatal MeHg exposure

The genetic predisposition of prenatal MeHg exposure on cognitive deficits (based on WISC-III) in 8-yr-old school children in Bristol, UK has been documented [174]. Also, a prospective study of 2-yr-old Taiwanese children suggested that variants of apoprotein E, a major protein transporter of mercury in the brain, may modify neurodevelopmental effects of MeHg [175]. Additional studies have confirmed that genetic factors can influence mercury toxicity [12,176-179]. Toxicogenomic is now being utilized as an analytical tool to detect mercury in seafood [180].

Prenatal MeHg exposure and heart disease

Regarding the question of whether prenatal MeHg exposure from fish consumption affects blood pressure in children, a study of 12 and 15 yr. old from the Seychelles Islands found no definitive correlation [181], However, similar 14 yr. study of the Faroe Islands birth cohorts observed decreased blood pressure attributable to MeHg neurotoxicity [182].

Several studies have concluded that MeHg from fish consumption may be a potential risk factor for CHD [22,183-186]. Such risk increases when the hair mercury concentration reaches $\leq 2 \ \mu g/g$ [187]. Nevertheless, no association between mercury level and risk of CHD was observed based on toenail analysis in the US, EU and Israel [188,189].

Two recent meta-analysis of fish consumption data noted 60g fish/d as the ideal dose for preventing CHD mortality [190]. Further, higher consumption of fish ($\geq 4x/wk$) has been associated with greater protection against CHD [191]. The daily fish intake of 60 g happens to be the average consumed by the Japanese who are the highest fish consumers in the world. An earlier analysis of 8 studies of fish consumers (>once/wk) found 17% reduction in CHD [192].

From consumption of MeHg contaminated other dietary constituents by mothers

Among the other dietary constituents besides fish, a Swedish study of pregnant women found that consumption of chicken increased cord blood MeHg levels [193]. Apparently, mercury in chicken originated from fishmeal used as chicken feed.

In high-rice consuming countries, such as China, India and Indonesia, rice is an important dietary contributor of MeHg [45,194,195]. Methylation of mercury in paddy soil, sediments and water used to grow rice plant appears to be the source of MeHg in rice [196]. In China, rice may contain \leq 569 µg of total mercury/kg, of which \leq 145 µg/kg is MeHg [197]. The Chinese rice contains higher mercury levels than those grown in other Asian countries [198]. Rice seed has high accumulation potential for MeHg but not for inorganic mercury [199,200]. Notably, there is no significant difference in MeHg concentrations in brown and white (polished) rice since it accumulates primarily in the endosperm [201].

Thus, in inland China, rice rather than fish is the major source of MeHg exposure [202,203]. Stable mercury isotopes (¹⁹⁹Hg and ²⁰²Hg) in scalp hair can be used to distinguish MeHg intake from rice *vs.* fish [204]. Compared to fish, rice has lower nutritional value and it lacks micronutrients, such as DHA and Se [205].

The recent survey of pregnant Chinese women found cord blood mercury levels (mean 2.26 μ g/L) lower than the WHO/ FAO recommended safe level (8.6 μ g/L) [206]. Also, the latest rural Chinese child neurodevelopment study (based on BSID-II MDI and PDI) reported that contemporary changes in family structure had impacted children's sensitivity to maternal MeHg exposure [202].

Incidentally, the Chinese international rice trade has significantly aggravated MeHg exposures in Africa (62%), Central Asia (98%) and Europe (42%) [27]. Additionally, vegetables, such as cabbage, celery, spinach, pumpkin and amaranthus [207-210], and mushrooms [211] grown in mercury-contaminated soil and water have increased dietary mercury contribution (>FAO/ WHO safe level). However, such mercury contamination can be mitigated by washing produce with water [212] or household vinegar [213].

From consumption of mercury contaminated fish and other dietary constituents by children

Ocean fish consumption is important for the health and development of children in many parts of the world [170,214].

The potential adverse effect of MeHg from contaminated fish consumption on children's IQ is being debated [156,215,216]. The amount of MeHg ingested depends on: 1) fish species, size, where and which season it was caught, 2) frequency of consumption, and 3) serving size (usually ~70 g/child) [129,130,217].

An advisory limit of 1 μ g MeHg/g fish has been set by FDA [218]. The EPA threshold mercury dose for neurodevelopmental effects from fish consumption is 0.1 μ g/kg bw/d [218]. Suggested weekly consumption quantities of DHA-rich fish (salmon, sardines, Atlantic mackerel, etc.) for children are: ages 5-11 yr. (average bw 14.4 kg), 40 g and ages 5-11 yr. (average bw 26.4 kg), 74 g with monthly limits for larger predatory fish (whales, shark, tuna, etc.) for 1-4 yr., 75 g and 5-11 yr., 125 g [219,220]. A recent survey of US children found blood mercury levels <EPA RfD (5.8 μ g/L) in 1-19 old [58] with no adverse effect on adaptive and problem behaviors (based on BASC-2 scores) [221].

The desirable mercury level in scalp hair of fish-consuming children is $<2.3 \ \mu g/g$ (WHO reference value) [222]. The form of fish consumed (fresh, frozen or canned) or how it is cooked (baked, grilled or pan/deep fried) generally does not affect its mercury content [223]. Since MeHg in fish has already bound to tissue proteins [73], it is not eliminated by cleaning or cooking [176]. However, acidic medium (*e.g.*, vinegar) can release mercury bound to fish proteins [224].

Based on VEP measurements in children, the Canadians have recommended the total blood mercury threshold limit of 8 μ g/L for pregnant and childbearing-age women [225,226]. Among the Inuit population of Nunavik in Northern Quebec, staple diet of fish and marine mammals has resulted in higher mercury levels in blood and scalp hair of their children, the former correlating with neuromotor functions [227-229]. Similar results have also been reported in the Arctic populations [25].

EU has adopted the EPA RfD for MeHg exposure [230]. Studies of mother/child pairs in Ireland found scalp hair mercury levels<1.0 μ g/g [231] and in the Italian coastal population of the Mediterranean Sea (which borders 21 countries), the MeHg levels from fish consumption were <TWI recommended by EFSA (1.3 μ g/kg bw) [232]. In contrast, children in Spain (Granada, Madrid, Murcia, the Ribera d'Ebre and Menorca Island) have higher levels of MeHg in scalp hair [233-235]. Their consumption of oily and canned fish positively correlated with cognitive functions (based on MSCA scores) and negatively with white fish (bass, catfish, cod, tilapia, etc.) intake, especially when fried.

The biomonitoring studies of mercury in the Japanese children found no adverse effects on neurodevelopment (BAEP test) or scholastic achievement (SACMEQ test) attributable to fish consumption [79,236]. The reference level for mercury in scalp hair in Japan, the world's largest fish consumer, is 3 μ g/g based on TWI of 2.0 μ g mercury/kg bw from fish.

Although Jamaican children consume large amounts of seafood and their blood mercury levels are much higher (0.99 μ g/L) than those in the US and Canadian children (0.33 and 0.31 μ g/L, respectively), no association between blood mercury levels and ASD was observed [237]. This paper discusses discrepancies in earlier studies that reported such association. A study with US children has also affirmed this conclusion [238].

The Chinese use fish congee as weening food for toddlers which may contain 50-520 μ g mercury/kg depending on the type of fish used. This can result in the child's weekly intake of 1.2-13.0 μ g mercury/kg bw. The infants fed such congee exceeding TWI of MeHg (1.6 μ g/kg bw) have exhibited neurological and other symptoms of mercury poisoning [239]. Also, the fish-consuming Chinese children in Hong Kong have blood mercury level >5.82 μ g/L with >9x higher risk of exhibiting ADHD [240].

In Australia, children's diet does not contain the recommended amounts of PUFAs, especially DHA since they consume 8.5x more meat than fish and seafood [241]. However, the high consumption of vegetables appears to help increase their total dietary PUFA intake [242,243]. Currently, infant formulas supplemented with DHA are popular in Australia [244]. The later-life benefits of DHA on neurodevelopment [245,246] and cardiovascular function [247] are the reasons for its supplementation.

The 2013 survey of 168 baby foods sold in the US found that 32% contained <0.146-4.060 μ g/kg of total mercury with those containing rice topping the list [248]. Also, marlin fish jerky snack food popular with children in Hawaii and California contains high concentrations of mercury (average 5.53 μ g/g) [249]. The recent US congressional report found up to 10 μ g/kg mercury in some US made baby foods [250]. FDA has proposed a 4-yr. "Closer to Zero Action Plan" for baby food toxic metal contaminants which includes mercury [251].

Mercury-contaminated fish consumption pattern around the world

Like rest of the world, fish consumption in the US is the primary way the women of childbearing age and their children are exposed to MeHg, and it is generally below the level of any health concern [252]. Even so, according to the NHNE survey of 1999-2000, >300,00 newborns each yr. may have been exposed in utero to MeHg levels >EPA RfD [253]. Also, high fish consumption among women of reproductive age, especially African and Asian Americans, has resulted in preterm births in Maryland, South Carolina, Louisiana and Florida [254-258] with potential lower childhood IQ [156].

The 2008-2009 EPA survey of 541 sites across the continental US found mean mercury concentrations of 21-1,419 μ g/kg fish [259]. The 2009-2012 survey of 5,656 US children ages 1-19 yr. found that 62.4% ate fish and had blood mercury levels below the EPA reference level, 5.8 μ g/L [260]. An earlier survey of 1-5 yr. old children had also reported the historically lowest blood MeHg and total mercury concentrations (0.17 and 0.26 μ g/L, respectively) [261]. However, the San Francisco Bay area children in California are known to be high-end fish consumers with elevated blood mercury levels [109]. Additionally, American children in Alaska and Hawaii have higher amounts of fish in their diet.

The fish consumption pattern in 17 EU countries has been studied [117]. The Portuguese are third largest fish consumers in the world, after Japan and Iceland. A survey of 343 pregnant women in Lisbon found their mean RI for MeHg exposure from

fish consumption to be 0.81 (calculated by fish intake in $\mu g/kg \text{ bw/d} \div 0.24 \mu g/kg \text{ bw/d}$, the WHO tolerable daily intake; the desired RI <1.0) [262]. Notably, the ingestion of black and silver scabbard fish enhanced mercury toxicity risk.

The 2015 Spanish consensus document on the prevention of MeHg exposure has recommended that pregnant and nursing women as well as children should consume fish containing mercury levels <0.15 mg/kg, (resulting blood levels of ~10.8 μ g MeHg/L and ~12 μ g total mercury/L) [217]. This is comparable to the Japanese (who consume more fish per capita) but higher than those in the US, Canada and other EU countries.

The fish consumption of Finnish mothers appears to compensate the benefit and risk to child's brain development based on IQ measurement [76]. Also, among the Finnish fishermen families whose intake of contaminated fish is high (64-89 g/d, 2x general population), no adverse effect on mortality is observed [263].

In South America, an analysis of 110 species of fish from Madeira River, the biggest tributary of Amazon in Brazil found a concentration range of 0.01-6.06 µg mercury/g [264]. Among the high fish-eating villagers (~406 g/d), mean scalp hairmercury level ($17.4 \mu g/g$) of breastfeeding mothers significantly correlated with their children's hair level [265,266]. While this had no significant impact on newborn birthweight [267], impairment in motor performance (TGMD-2 test) was observed at ages 7-11 yrs. [268]. However, such developmental delays were also attributable to their health inequalities and socioeconomic disadvantages [96].

The children with fish as the main component of their diet (283 g/day containing $\leq 0.2 \ \mu g$ mercury/g) in the fishing communities along the Caribbean coast of Colombia are potentially exposed to MeHg 3x WHO/FAO TWI, 1.6 μ g/kg bw [269]. Similarly, the children living in the northern border area are exposed to high levels of MeHg from consumption of fish from rivers polluted by mercury-containing wastewater from the artisanal gold mines [270].

West Bengal is a high fish consuming state in India with a population of >10 million children of age \leq 6 yr. The 2016 analysis of scalp hair of younger residents (<21 yr.) of a fishing community near Kolkata, the state capital found mean total mercury value less than EPA RfD (1.0 µg/g) [271]. The city residents had lower hair mercury level (0.49 µg/g) compared to those from the fishing community (0.83 µg/g), presumably due to lower fish consumption. Also, recent surveys of commercial fish from the Bay of Bengal [272], coastal Mumbai [273] and Goa, another high fish consuming region [274], found mercury levels within the permissible limits.

Over the past five decades (1961-2011), there has been a worldwide increase in human exposure to MeHg from fish consumption (>TWI 1.6 μ g/kg bw) [275]. The annual health benefits of a 10% reduction in MeHg exposure in the fish-consuming US population is estimated to be \$860 million [276]. Of this, 80% is associated with reduction in fatal heart attacks and 20% in IQ gains in children. EFSA [277] has recommended that each country needs to consider its own pattern of fish consumption for risk-benefit analysis. Sushi fans should note that raw fish may contain 55%-60% more bioaccessible mercury than in cooked or fried ones [278,279]. Reports of biomonitoring and risk assessment of mercury contamination in fish from various parts of the world are available [280-282]. Also, there is a compilation of mercury content of vegetables, fruits and fish consumed in India which, in general, is within TWI [283].

A field experiment in France found that risk-benefit advisory had minimal effect on consumer fish choice [284]. Instead, store warning labels on fish with high mercury content were recommended as a more effective tool. In the US, consumer advisories also have no major impact on fish choices made by the women of childbearing age [285-287].

Thus, both maternal and children's consumption of mercury contaminated fish and other dietary constituents continue to be of active research interest around the world.

Mercury vapor exposure from mercury-containing devices

Mercury is the only element that is liquid at room temperature, 13.6x heavier than water and readily evaporates. One 4-mm diameter bead of mercury (0.034 ml or 0.46 g) can generate at 0.1 m above ground, up to 0.56 μ g mercury vapor/m³ of room air in 30 min at 170°C. This vaporization increases rapidly as temperature rises, *e.g.*, ~6x at 38°C [288].

Since mercury vapor is colorless and odorless, its detection in home could be challenging without professional help. If concentrations >1 µg mercury/m³ room air are detected, cleanup should be initiated and residents evaluated for exposure [289]. Their relocation is called for when the mercury reaches toxic level of $\ge 10 \text{ µg/m}^3$ room air [290-292].

The most common cause of mercury vapor exposure of children at home is due to broken fever thermometers, blood pressure monitors and light bulbs. Although spills from broken thermometers (which contain 0.5-0.7 g mercury) rarely reach mercury vapor levels >1 μ g/m³ room air [43,64,292]. they could create hazardous conditions to infants if such indoor spills are improperly cleaned (See Clinical signs, etc. section below). In contrast, blood pressure monitors contain larger amount of mercury (~150 g) and when broken, they are more likely to create a hazardous situation, as in the case of a home day care center in Hillsborough County, Florida. Unknowingly it utilized a leaking antique monitor (resulting in ≤ 89 μ g mercury/m³ room air) as an educational toy [293]. Also, a recent mercury spill entered HVAC system in the basement of a house in Virginia exposing a family with 3 children to toxic vapors [294].

Since spilled mercury disperses into small droplets that get embedded in carpet fibers and floor cracks, vacuum cleaning produces aerosols and enhances its vaporization. Thus, the recent attempt to vacuum clean a spill of ~40 g of mercury from a broken barometer in a Netherland home resulted in high blood levels (26-32 μ g/L) in a boy (9 mos.) and his sister (2.5 yrs.) within 6 h of exposure [295]. Similar poisoning of children in the US attributed to vacuum cleaning of mercury spills from broken thermometers has been reported [67]. Hence, vacuuming or using broom to clean mercury spill is not recommended [289]. Poor ventilation and elevated temperatures further increase mercury levels in the room air. Activated alumina may be used to cover-up inaccessible mercury spillage to reduce vaporization [9]. Mercury spills larger than that from a broken thermometer need to be promptly remediated by professionals [296,297]. A review of health consequences of mercury spills from common devices at home is available [44]. Most countries have mercury emergency phone hot lines. In the US, it is 800-220-1222 at the Poison Control Center.

In contrast, the widely-used 4-ft long fluorescent tube lights contain smaller amounts of elemental mercury vapor (12-20 mg) and only 6% of this is released to the air when broken. Notably, mercury binds to glass as the bulb ages and ~4 mg is oxidized [298]. The compact FLBs contain even smaller amounts of mercury vapor (3.0-4.5 mg). Thus, mercury hazard from broken light bulbs in homes is minimal. In most instances, the small amount of mercury released can be adequately vented by opening windows and using exhaust fans to achieve the EPA reference level ($\leq 0.3 \ \mu g/m^3$) [289].

In the rare instances of broken fever thermometers in the mouth of young children, as recently reported in Shanghai, China, X-ray images can help locate mercury residue and in one child, local excision was resorted to remove it from the floor of mouth [299]. Accidental ingestion of mercury from a broken thermometer is generally non-hazardous due to its poor absorption in the oral cavity and GI tract [8]. There is a recent report of mercury in the vomitus after drinking milk spiked with the metal [300]. Similarly, absorption of elemental mercury through skin contact is low (0.024 ng/m² for each mg/m³ room air) [61].

Exposure from heating mercury to extract gold and silver from ores and scrap

Extraction of gold and silver from ores by mercury amalgamation and from scrap dental fillings involves heating which could generate potentially lethal concentrations of vaporized mercury (0.193-0.370 mg/m³ room air). Such operations at home by amateurs, especially in the poorly ventilated residential kitchens have resulted in children's death [301,302]. Also, fetal uptake of mercury could take place in exposed pregnant women due to its facile transfer across the placenta [303]. (See Clinical signs, etc. section below).

To avid entrepreneurs, there is good news of the availability of a mercury-free gold extraction procedure which uses Borax (sodium borate), a common ingredient of household cleaners and laundry detergents [304].

Exposure from cultural or religious ceremonial uses of mercury

Some Caribbean and Latin American religions, such as Voodoo, Santeria, Palo and Espiritismo use mercury ceremonially and apply it to the skin, add to candles or sprinkle around the house [46,55,303,305]. As a precaution, homes of these religious practitioners should be monitored to assure mercury levels are <1.0 μ g/m³ room air. Also, blood, urine and scalp hair levels of mercury should be checked in young children suspected of such ritualistic exposures [306].

Exposure from unregulated cosmetics, herbal medicines, dietary supplements, toys and jewelry

Chloride and bromide salts of mercury are commonly used in cosmetics, herbal medicines and dietary supplements [307].

The 2003 FDA list of mercury containing medicinal products is still of current relevance due to their continued popularity [308]. Also, low-cost jewelry and toys from Mexico and Asian countries are often contaminated with mercury [309-311].

Parents and other family members use of facial skin-lightening creams containing mercury salts (which inhibit melanin formation) may expose children to the toxic metal [66,312-316]. Also, mercurous chloride (calomel) in teething powder is known to cause childhood mercury poisoning (acrodynia or pink disease, see Clinical signs, etc. section below).

Herbal medications sold OTC in pharmacies and *via* the internet are often contaminated with mercury (some as high as 103 mg/g). Also, mercury concentrations >4.2 mg/g in herbal dietary supplements are not uncommon, with bamboo shoots and green microalga being the frequent contributing ingredients [317,318]. However, mercury in traditional Ayurvedic medicinal products from India is generally not a contaminant but added as an active ingredient [319]. Mercury poisoning of children in the US and EU by such medications has been reported [320].

Moreover, herbal teas could be a significant source of mercury exposure in children [321]. (AAP recommends not to feed infants herbal teas (which may also contain other toxins, such as pennyroyal oil in mint tea) which could lead to fatalities [322].

In the US, FDA (per 21CFR700.13) has approved the use of mercury compounds as preservative in eye products only (at \leq 65 µg/g) [308]. Thus, all cosmetics containing mercury >1 µg/g. with the exception of eye products, are considered contaminated and subject to regulatory action.

Exposure from workplace tracking of mercury to homes

Elevated levels of mercury in children and homes of workers of plants manufacturing thermometer, FLB and chlor-alkali products have been reported [299,323,324]. Thus, in the US, the urine mercury level was higher (25 μ g/L) in the children of a Vermont thermometer plant workers than those of non-mercury plant workers (5 μ g/L) [324]. Also, mercury contamination in homes of a chlor-alkali plant workers in Charleston, Tennessee during scheduled maintenance has been reported [323]. Although no toxic effects were observed in both cases, the children of mercury plant workers are potentially at risk and monitoring is advised. Additionally, it is prudent for the plant workers to wear separate work clothes and shoes to prevent carrying mercury contamination outside of the work area [9].

Workplace tracking of mercury from clinic to home by dental professionals who work with mercury containing amalgam fillings is considered not significant because of the relatively small quantities of the metal they handle. (See Exposure to mercury used in dentistry section below).

A recent indoor air mercury monitoring in ten hospitals in Bali, Indonesia found that 90% of the hospital area had <1 μ g/m³, 9% 1-10 μ g/m³ and 1% >10 μ g/m³ with higher concentrations in emergency rooms and dental clinics, and the highest in equipment repair/maintenance workshops [325].

Exposure from industries that use mercury near homes

Chlor-alkali plants produce chlorine, hydrochloric acid, caustic soda (sodium hydroxide) etc., using mercury cells each

containing about 8,000 lb of mercury. A typical plant uses about 56 such cells [326]. Elevated levels of MeHg in water (7 ng/L) and fish (5.2 mg/kg) in a Romanian reservoir due to microbial methylation of mercury released from a chlor-alkali plant have been documented [327]. Similarly, leafy vegetables grown near a chlor-alkali plant in Ganjam, Odisha State, India contained elevated mercury levels (8.9 mg/kg) [328]. Also, higher atmospheric levels of mercury (27.4 μ g/m³ air) in the vicinity of chlor-alkali plant in Flix, Spain are reported [329]. The scalp hair MeHg analysis of 4-yr. old children living near this plant found twice the amount (0.631 μ g/g) compared to those not living near the plant [330]. However, their hair mercury levels decreased over the 13-yr. period with no correlation with neurophysiological test scores (TMT-B and FTT) and ADHD [331]. Similar results were obtained in the vicinity of a chloralkali plant in Portugal shut down after 50 yr. of operation [332]. Further, a detailed study of ambient air, soil and vegetables grown in the vicinity of a chlor-alkali plant in Tuscany, Italy found mercury concentrations within the EU safe level [333].

In 2014, there were an estimated 50 chlor-alkali plants around the world and in the US, just 2 as of 2018 [334]. The environmental exposure hazards to children are minimal from other industries that use smaller quantities of mercury (*e.g.*, thermometer and FLB manufacturers). An exception was a US-owned thermometer factory in Kodaikanal, Tamil Nadu, India. It was shut down for blatantly polluting the pristine environment of the popular hill station by discharging waste mercury in the early 2000s [335].

Mercury exposure at schools

Students are attracted to silvery liquid mercury which disperses into tiny droplets and quickly forms large globs upon shaking or scooping with fingers [336]. The common sources of mercury vapor exposure at school are: 1) elemental mercury stored in science laboratories, 2) mercury from broken instruments and FLBs, and 3) gymnasium floors covered with certain polyurethane material (such as 3M Tartan) manufactured prior to 1985 using mercury-containing catalyst [61,310].

Student misuse of mercury accounts for numerous short-term exposures to its vapor as reported in Arizona, Mississippi, Missouri, Nevada, Texas and Washington, DC schools in the US [68,337-339]. Thus, mercury stolen from storage rooms was taken to class rooms, gym and homes. The air mercury levels measured were highest near the student locker rooms ($50 \mu g/m^3$ compared to the background, $0.01-0.04 \mu g/m^3$). The mean urine mercury level of 200 students tested was $0.36 \mu g/L$ (range $0.14-11.4 \mu g/L$) with higher levels in those touched mercury and/ or got it on their clothes. One school was closed for 35 days for cleanup and over 200 homes were tested for contamination. In the most recent incident, a high school in Chicago, Illinois was evacuated after the discovery of mercury in bathroom toilet (Chicago Tribune, Jan 14, 2023).

In general, there have been not many reports of severe adverse effects in students that required medical attention due to mercury exposures in the US schools [337]. In 2020, three students in Dallas, Texas developed symptomatic elemental mercury poisoning that required hospitalization and chelation therapy with DMSA [68].

Besides the US, in the past 10 yrs. many students in Turkey were poisoned by mercury in several schools [69,340-342]. Over 250 children were exposed by unauthorized handling of mercury and in one case, 26 were intoxicated as the result of a broken mercury thermometer in a hot, closed-door laboratory. These acute mercury vapor exposures were detected early on and successfully treated with D-penicillamine or N-acetyl cysteine.

A mercury awareness guide for school teachers is available [343]. The acceptable level of mercury in indoor air for school is 1-3 μ g/m³ which is higher than that for home (<1 μ g/m³) accounting for less time spent in school by children [344].

Mercury exposure at other locations

In the US, children's exposures to mercury vapor at other locations, such as repurposed daycare facility at a former mercury thermometer factory [345] and residential condominium conversion of a building that used to manufacture mercury vapor-containing light bulbs, both in New Jersey, have resulted in enhanced urine mercury levels due to inadequate remediation [346].

Also, there are reports of children's exposure to mercury discarded on abandoned industrial properties. Thus, two teenagers scavenged a large amount of mercury (~23-100 lb) from a shuttered neon sign factory in Texarkana, Arkansas and contaminated 12 residences, 1 convenience store, and a local school [347]. One home and an apartment were so severely contaminated that both had to be demolished. Although exposed children had high mercury levels in urine (66.6 μ g/g creatinine) and blood (104 μ g/L), no lasting adverse effects were reported because of prompt intervention.

Similar incident of youngsters hoarding large quantities of mercury (~220 lb which needed a wheelbarrow to haul) from a railway tilt switch facility in Manchester, UK has led to exposure of 225 juveniles [348]. Many were at high risk and needed chelation therapy with DMPS. Also, in 1993, children in Hamilton, Canada took mercury from an abandoned metal recycling plant and distributed to their peers creating a major mercury emergency [349]. This resulted in screening of ~6,000 children, and 269 were identified as exposed to mercury. Fortunately, none of the exposed children exhibited adverse health effects thanks to quick action taken by the local public health officials.

Other examples of innocuous mercury vapor exposure events include mercury poisoning of a 24-mo.-old toddler in Syracuse, New York whose case of acrodynia (peeling pink skin at the tips of fingers and toes) was due to mercury vapor from broken FLBs stored in a shed next to his nursery [350] and mishandling of the mercury spillage from a broken blood pressure monitor in the waiting room of a clinic in Detroit, Michigan [351]. In the latter case, no child was harmed thanks to the prompt response by the alert staff.

Mercury exposure from other sources

The US EPA [334] list of the major contributors to mercury pollution of the air includes: Chlor-alkali plants, coal burning power plants, waste disposal and landfill sites, auto crushing yards (mercury from antilock brakes), recycling facilities for FLBs, thermometers, switches and gears, human crematoriums (mercury released from dental amalgam fillings) and mobile sources (locomotives and marine vessels). It is noteworthy that adverse pregnancy outcomes in communities near incinerators and crematoriums have been reported [352].

The sobering finding of 80% reduction in atmospheric mercury pollution from 1990 to 2014 (~250 to ~50 tons) [334] is attributed to both reduced mercury usage due to the development of mercury-free new technologies and implementation of better pollution control measures. Reviews of recent abatement innovations in industrial mercury pollution are available [353,354].

The total amount of mercury released to environment from global coal combustion is estimated to be 38 Gg (~71% atmosphere, ~31% land and water) [355]. Most of it is released in Asia and Europe (32% each). A 2005 analysis in the US concluded that unilateral reduction of mercury emissions from coal-fired power plants alone is unlikely to realize any significant public health benefits [356]. Less than 2% of the total mercury emitted from 3 coal-fired US plants was detected within 15 km (9.3 miles) of the plants with low health risks attributable to such emissions [357,358]. While soil mercury levels were low and considered safe, there is concern for exposure risk to toddlers playing out doors due to potential oral contact [359].

Interestingly, lower tissue mercury and higher Se concentrations were found in fish from lakes near power plants [360]. Also, a survey of 2-yr.-old Chinese toddlers with prenatal exposure to mercury in coal-burning plant pollutants found no development defects based on GDS [361]. However, a recent study has reported higher mercury levels (>WHO TWI) in grain and vegetables grown near coal-burning power plants [212].

The cement industry in India is the second largest in the world (next to China) and emits 45.6 mg mercury/Mg cement produced. This is substantially less than the EPA reference level of 65 mg mercury/Mg [362]. While the limestone raw materials used are the source of such contamination, both India and China have yet to set mercury emission standard for cement plants [363]. Studies have shown that cement plants located in urban areas can increase health risk of children. Thus, a survey of school children in Barletta, Italy found elevated toenail mercury level (0.15 μ g/g vs. 0.09 μ g/g in control group) attributable to a nearby cement plant emission [364].

Following the unprecedented terrorist attack on World Trade Center in New York City by crashing two highjacked passenger planes into its 110-story twin towers on September 11, 2001, large amount of mercury (from thousands of gallons of burning jet fuel and gasoline from parked automobiles, their exploding antilock brakes and FLBs in the two towers) was released into the air. While blood mercury (2.29 μ g/L) was not significantly raised in pregnant women living or working near the crash site, higher cord blood mercury (5.0 μ g/L) was associated with reductions in developmental scores (MDI and PDI) in their children [365].

Historically, amalgamation with mercury has been used for more than 4,500 years in mining precious metals [366]. A literature review and bibliometric analysis of the current use of mercury in gold mining is available [367]. About 37% of mercury emitted to global environment is produced by ASGM regions located in some 19 countries in South America, Africa and Asia [368]. The scalp hair and urine mercury concentrations of children in these gold mining regions are>WHO guidance value and correlate with lower neurophysiological test scores (SBIS and WISC-III), and kidney and immune dysfunctions [368-372]. But such mercury exposure has decreased in recent years due to diminished mining activity as alluvial deposits got depleted [373]. However, in some ASGM regions children continue to be exposed to higher levels of mercury [10] exacerbated by consumption of heavily contaminated fish, *e.g.*, Peru, Senegal, French Guiana and Ivory Coast [111,366-377]. (See Simultaneous exposure.... section below). Also, Mexican children are exposed to higher levels of environmental mercury pollutant from industries performing primary and secondary extraction of gold and silver [378].

Among the current productive cinnabar (mercuric sulfide, the most common mercury ore) mining regions, the following scalp hair mercury levels in children are reported: 1.4 μ g/g in Wanshan, Guizhou, the largest in China [379] and 2.64 μ g/g in Almaden, Spain, the site of one of the world's oldest and largest mercury deposits [380].

Exposure to mercury used in dentistry

Ancient Egyptians were known to use a mercury amalgam to fill tooth cavity over 1,500 years ago [35]. First documented in a Tang Dynasty Chinese medical text by Su Gong in 659, the widespread use of mercury-based dental amalgam is attributed to the invention of the English chemist Charles Bell in 1819 [381]. A typical dental amalgam contains 50% mercury, 30% silver and the remaining 20% copper, tin and zinc. Worldwide it is estimated that 100 tons of dental amalgam mercury enters the waste stream annually [13].

The three potential sources of children's exposure to mercury from dental amalgam are: 1) occupational exposure of their parents in dental profession, 2) systemic pre-and postnatal release of mercury from dental amalgam fillings in mothers, and 3) mercury released from dental fillings in children themselves.

From exposure of parents in dental profession

In the dental office, dentists and their staff are exposed to mercury vapor while filling tooth cavities with mercury-based amalgams [13] and during sterilization of amalgam-contaminated instruments [382]. Potentially harmful urinary mercury levels (20-50 µg/L) in the dental personnel were common during 1960-1970's [9,383]. The modern dental practice complies with NIOSH safe mercury exposure level (≤ 0.05 mg/m³ clinic air). This has been affirmed by the 35-yr. (1974-2009) mercury monitoring of dental clinics [384]. Further, recent studies of pregnancy outcomes among dental professionals found no increased occurrence of birth defects attributable to prenatal occupational exposure to mercury [385,386].

From release of mercury from dental amalgam fillings in mothers

At body temperature of 370°C, the average amount of mercury vapor emitted in the mouth from the surface of dental amalgam filling is 1.2 μ g/cm²/d, and its systemic uptake and urinary excretion are well-documented [9,61,387]. In the mouth, mercury is methylated to form MeHg by the oral bacteria and

accounts for its non-dietary source in the body [388]. For every 10 amalgams placed in the mouth, the urinary mercury level increases by 1 μ g/L [389].

Mercury released from maternal dental amalgam fillings passes on to fetus and breast milk, but unlike MeHg from fish, there is no definitive evidence for its adverse health effects [55,390-392]. A survey of Brazilian nursing mothers found no correlation between breast-milk mercury and dental amalgam fillings [393]. Also, the Norwegian mother/child cohort study of children born in 1999-2008 found no unequivocal correlation between number of dental amalgam fillings and perinatal death [394].

From release of mercury from dental amalgam fillings in children

Like adults, children are also susceptible to mercury vapor exposure from dental amalgam fillings [395]. As with adults, there are no reports of adverse health effects of such mercury exposure in children [396,397]. Thus, a study of Portuguese children in Lisbon found no behavioral effects of mercury amalgam fillings when compared with mercury-free composite resins used to fill dental cavities [398]. Also, the children in 17 EU countries showed no significant contribution to scalp hair mercury level from dental amalgams [117]. However, the number of amalgam fillings had a significant dose-response relationship with urine mercury levels which increased with gum chewing [399].

The FDA White Paper has concluded that both maternal and children's dental fillings pose no health risk to children [45]. It noted that neurologic and organ-specific effects appear when mercury vapor levels reach >50-100 μ g/m³ air or >50-100 μ g/g of creatinine in urine. The best estimates of mercury exposure in children from all sources are substantially less than these reference toxic levels [400]. A recent study of Spanish children found that mercury neurotoxicity is not associated with ASD [401]. However, the topic of mercury exposure and ASD continues to be of active interest [41,402-404].

The American Dental Association has recently updated its longstanding affirmative position on dental amalgam safety with the statement: Although amalgam remains an effective and inexpensive restoration option, environmental concerns regarding mercury have fueled legislative and regulatory actions in other countries to phase down amalgam use [405]. Denmark, Norway and Sweden have banned the use of mercury amalgam fillings and its use is severely restricted in Japan [406]. Also, as of 2018, UK and EU no longer allow mercury amalgam use in children under the age of 15 yr. and a recommendation for its complete phase-out by 2030 is being considered [407]. Further, Germany and Canada advise against its use in children and pregnant women [408].

There is encouraging news for parents concerned about the safety of dental amalgam containing mercury. Silver diamine fluoride appears to be a viable alternative to mercury amalgam for filling cavities in children which requires no anesthesia or drilling [409]. Although this mercury-free restorative compound has shorter durability (~5 yr. compared to 15-20 yr. for mercury amalgam), it is suitable for filling deciduous teeth cavities in children.

Simultaneous exposure of children to multiple forms of mercury

Children are simultaneously exposed to multiple forms of mercury: 1) elemental mercury vapor in the environment, contaminated home and from dental amalgam, 2) MeHg from breast milk, fish and seafood, 3) ethylmercury from vaccines, and 4) inorganic mercury from family use of unregulated cosmetics, dietary supplements and herbal medicines.

A Swedish longitudinal study found distinct levels of intoxication for each mercury species (metallic, inorganic and organic) with dominant contributions from maternal MeHg from fish diet and elemental mercury from dental amalgam fillings [410]. The serum and scalp hair mercury levels depend on the type of fish consumed, Se content in the diet, the number of amalgam fillings and where the children reside [411,412].

The New Hampshire Birth Cohort Study using maternal toenail analysis found that gestational mercury exposure from seafood consumption and dental amalgams was associated with increased risk of lower respiratory infections in infants [110]. However, the British Avon longitudinal study found that such prenatal exposures, in addition to ambient air mercury were not adversely associated with offspring IQ (WISC-III) [413]. Also, the atmospheric mercury exposure of Chinese children from fossil fuel combustion coupled with MeHg intake from rice and fish diet generally resulted in low cord blood [414] and scalp hair mercury levels [190,194,207,415].

In contrast, a comprehensive review of 72 studies of children living in 19 South American, African and Asian ASGM communities exposed to mercury in the environment and MeHg from fish diet found: 1) high mercury concentrations in their scalp hair (>2.3 μ g/g) and urine (up to 667 μ g/g creatinine) [368] and 2) they were susceptible to kidney and neurologic toxicities [1]. (See Mercury exposure from other sources section above). An additional source of mercury exposure is ethylmercurybased preservative used in children's vaccines [308]. Consequently, children in such ASGM communities continue to be exposed to higher levels of mercury [10,375]. While several epidemiological studies have found no conclusive evidence for neurophysiological or kidney toxicities attributable to such mercury exposures in infants and young children [55,416,417], the clinical manifestations may be delayed due to the latency period [78,80,418,419] and follow-up studies are warranted.

Thus, when evaluating mercury poisoning of children, simultaneous exposures to its different forms from all sources need to be considered.

Clinical signs, symptoms and treatment of mercury poisoning of children

The medical journals around the world are replete with articles describing the clinical signs and symptoms of mercury poisoning in children. Chelation therapy appears to be the most frequently used treatment for mercury poisoning.

General toxic effects

Environmental mercury exposure affects anatomical, physiological, metabolic and functional processes in children involving CNS (brain and spinal cord), lungs, kidneys,

liver, eyes, ears, skin, and digestive and immune systems [1,15,22,56,65,419]. The symptoms and severity of mercury toxicity vary with its form (elemental, organic or inorganic), exposure mode (acute or chronic), route (inhalation, oral ingestion or skin absorption) and dose. For example, effects of different forms of mercury on eyes are: 1) elemental mercury vapor causes inflammation of cornea (keratitis by calcium deposition), 2) organic mercury impairs hearing (by damaging auditory hair cells) and vision (by constricting visual fields and degrading VER), and 3) inorganic mercury damages corneal endothelium (by changing opacity) and lens (by brown discoloration known as mercurialentis, a trademark early sign of mercury poisoning) [342,420].

Organic mercury compounds cause CNS, neurological and behavioral adverse effects similar to elemental mercury vapor exposure, *e.g.*, seizures, bronchitis, pneumonia, loss of motor and cognitive skills [49,103,342,421-424]. Other toxic signs of mercury vapor inhalation common with skin contact with its inorganic salts are: irritability, stomatitis (inflammation of oral mucous membranes), erythema (abnormal redness of hand, feet and other body parts), acrodynia, and erethism (behavioral and personality changes, extreme shyness, excitability, loss of memory and insomnia). Oral ingestion of inorganic mercury compounds causes GI distress and kidney damage.

The most frequently reported non-neurologic adverse effects of mercury and its compounds in children are contact dermatitis and cutaneous poisoning [8,34,425]. The hematological effects (anemia, lymphopenia, etc.) of mercury exposure appear to be rare in children [426].

Since mercury is in ionic form (Hg+ or Hg++) in its salts, they are readily absorbed in the body. In contrast, elemental mercury is slowly absorbed due to its nonionic form (Hg^o) [8,427]. Thus, orally ingested liquid mercury passes in the feces unchanged with little absorption or discernable toxic effect.

In general, among the three forms of mercury, chronic exposure to organic mercury compounds is more toxic to children than its metallic or inorganic forms. Conversely, acute exposures to mercury vapor and its inorganic salts are more toxic to children than organic mercury compounds. However, all three forms can be fatal in high doses. The major differences between organic and inorganic mercury poisoning are: MeHg causes brain damage, while mercury vapor damages lungs and mercury salts impair kidneys. The pathological differences in poisoning due to the three forms of mercury are well documented [426].

The potential toxic effects of maternal mercury exposure in children are: 1) prenatally: miscarriage and stillbirth, 2) at birth: low body weight, congenital malformations, vision and hearing deficiencies, and cognitive dysfunctions, 3) at infancy and childhood: higher mortality, asthma, neurobehavioral and immune impairments, 4) at adolescence: precocious or delayed puberty, and 5) at adulthood: increased risk of cancer and heart diseases [74].

Biomarkers and biomonitoring of mercury exposure

The frequently used biological specimens as biomarkers of mercury exposure in children are: blood, urine and maternal breast milk (which represent recent exposures), and scalp hair and nails (which represent long-term exposures) [83,287,395,418,428-434]. Both blood and urine may be used as bioindicators of MeHg and inorganic mercury exposures from fish consumption, the latter includes mercury formed by demethylation of MeHg in the body [435]. Also, mercury level in merconium (first stool) may be used as a convenient biomarker for evaluating antenatal exposure [118].

While blood mercury level does not correlate with urine level, both correlate with scalp hair level which offers a noninvasive, easy method of sample collection in children [436,437]. Using an average growth rate of 1.1 cm/mo. for scalp hair, a hair to blood ratio of 250 is an acceptable average value for safe mercury exposure [438]. MeHg in cord blood (for fetal exposure), and in hair and plasma (together with total mercury) are suitable biomarkers of mercury exposure in children [433,434,439]. Also, cord blood MeHg level significantly correlates with maternal hair mercury level [79].

Single blood mercury data should be treated with caution since it is not an indicator of long-term exposure [440]. Exhaled breath and sweat levels of mercury are important to measure since both could be significant routes for its elimination from the body [441]. However, there is no correlation between these two routes of mercury elimination and by the urinary pathway [442,443].

Total mercury levels in finger and toe nails correlate with scalp hair level attributable to fish consumption [1,430,444]. Further, hair from pubic and other regions of the body is also suitable for mercury analysis and the results correlate with scalp hair level [384,445,446]. Mercury concentration in saliva of children is generally below the limit of quantification. Thus, it is not suitable for monitoring mercury exposure [447].

An analysis of serum immune biomarkers in the fish-consuming teenagers from Amazonian Brazil found a positive association between mercury exposure and antinuclear antibodies and cytokines [448]. The results were similar to occupational exposure to elemental mercury vapor in adults (*e.g.*, ASGM workers) with symptoms of autoimmune dysfunction and systemic inflammation [449]. These serum biomarkers along with urinary ketoisoproporphyrin may be used to monitor long-term risks of prenatal mercury exposure in children [448,450,451].

In the US, CDC [452] and EPA [453] utilize biomonitoring of mercury in fish and other foods along with the environment to assess children's exposure to mercury. Such monitoring in New York City has identified skin care products as a source of inorganic mercury [316]. Similar biomonitoring of mercury is also being conducted in EU [3,114,230,231,281,433,447,454] . Japan [79,235] and India [280]. On a global basis, WHO, UN and other health organizations are conducting biomonitoring of mercury exposure of children in African, Asian and South American countries as well [6,60,271,373,448].

The mercury biomonitoring studies provide reference ranges to help determine whether children are exposed to higher (toxic) doses [132,252,260,261]. Additionally, such data are useful to evaluate the potential risks of prenatal environmental mercury exposure in different regions of the world [450].

Clinical analysis of mercury and chelation therapy

Cold vapor atomic absorption spectroscopy is a convenient laboratory method for the analysis of mercury in biological samples (detection limit 1 ng/L) [455]. A compilation of health-based guidance values for total mercury and MeHg in blood, urine and scalp hair of children and pregnant women is available [6]. While there is variation in the data from different countries, they are useful for comparative purposes [47]. Also, reviews of the low-dose mercury exposures and children's health [54,456], and guidance for identifying those at risk [60] are useful for reference. Interestingly, in recent years China, the most populous country in the world has reported relatively few cases of mercury poisoning of children [65].

Interpretation of results in children is relatively straightforward when biomarker mercury levels are highly elevated, but it becomes increasingly difficult as they approach normal values (e.g., blood and urine mercury <10-20 µg/L) [344,418,457]. While diagnosing mercury poisoning in children can be challenging [458], it can be made with reasonable reliability and successfully treated with chelation therapy [34,459,460]. Although DMSA (succimer) is approved by FDA as the chelator for use in children [67,461], DMPS [348], N-acetyl cysteine [341], D-penicillamine and its N-acetyl derivative [34,462] can also be used effectively. Chelation therapy increases mercury excretion from the body and the common side effects are: abdominal cramps, drowsiness, dizziness, rash, pruritus (itchy skin), and flu-like symptoms [418]. Evaluation of chelation treatment with Se supplementation [463] and potential inappropriate use of chelating agents [464] are noteworthy.

Specific clinical signs and symptoms of acute and chronic poisoning by the three forms of mercury in children along with successful treatments employed are described below.

From exposure to elemental mercury vapor

As indicated earlier, home is the most common site for accidental inhalation of elemental mercury vapor by children (See Mercury exposure at homes section above).

From acute exposure to low to moderate doses of mercury vapor

Acute (short-term) exposure of children at low to moderate doses of elemental mercury vapor ($\leq 1 \ \mu g/m^3$ air) generally results in headache, dizziness, insomnia, dilated pupils with vision defects, peripheral neuropathy, and involuntary movements. However, fatality due to such acute exposure is unlikely.

Accidental small spill from broken fever thermometer in homes rarely results in mercury levels >1 μ g/m³ room air [61]. (Mercury vapor exposure from mercury-containing devices section above). Nevertheless, a recent mercury poisoning case mimicking an infectious disease in three siblings in an Atlanta, GA home has been attributed to mercury spill from a broken thermometer exasperated by repeated vacuum cleanings. The resulting blood mercury levels (>200 μ g/L) needed chelation therapy with DMSA [67].

Also, school chemistry laboratories are potential sites for student exposure to low to moderate levels of mercury vapor (See Mercury Exposure at Schools section above).

From acute exposure to high doses of mercury vapor

The common symptoms of inhalation exposure to high doses of mercury vapor (>5-50 μ g/m³) are: salivation, swollen gingiva, fever, dry cough, dyspnea (shortness of breath), abdominal pain, nausea, vomiting and diarrhea [427], and death due to pulmonary dysfunction [55].

Although the exact fatal dose of mercury vapor is not known, exposure to >1-2 mg/m³ air for a few hours can cause acute chemical bronchitis and pneumonitis [465]. Lung injury appears as hyaline membrane formation two hours after exposure, followed by extensive pulmonary fibrosis.

Just how quickly inhalation of high doses of mercury vapor can lead to death in infants is exemplified by the recent fatal poisoning of a 14-mo. girl in Denizli, Turkey [466]. Her sister had brought home mercury from the school without permission from teacher. After playing with it, she placed the liquid metal on heating stove and watched it vaporize. A day later, her baby sister got fever and died. The autopsy indicated necrotizing bronchitis, pneumonia or respiratory distress syndrome attributable to high mercury vapor exposure with death by cardiorespiratory collapse.

Similarly, exposure to high mercury vapor concentration of 0.193 mg/m³ in kitchen air at a home in Fresno, CA, resulting from gold ore processing, has led to death of a 13-mo boy and his 38-yr mother [301]. Both had developed fatal respiratory distress within 48 h with blood mercury levels reaching 160 and 322 μ g/L., respectively. Death of a 7-mo. infant under similar circumstances has also been reported earlier in Springfield, MA [302].

On a positive note, there is a report of accidental inhalation of mercury vapor of very high concentration (~0.37 mg/m³ room air, produced by the home owner heating mercury-gold amalgam over a kitchen stove to recover gold) by a 19-yr. pregnant woman in Canada who 26 d later delivered a healthy infant with no detectable clinical abnormalities [303]. She was admitted to the hospital after 6 h of mercury exposure with paroxysmal cough, dyspnea, chest pain, nausea and vomiting. Three other children, ages 16 mos. to 7 yr., residing in the same bungalow were also treated for mercury poisoning with similar symptoms. Fortunately, in this case, exposed mother, newborn and all three children fully recovered after chelation therapy with D-penicillamine. Similarly, a 3-mo. infant in Shanghai, China was poisoned by inhalation of high concentration of mercury vapor which resulted in pneumothorax (collapsed lung) and respiratory failure 2d after exposure. She was successfully treated with DMPS chelation therapy [467].

From chronic exposure to low, moderate or high doses of mercury vapor

Children chronically exposed (3-4 mos.) to low levels of mercury vapor ($\leq 5 \ \mu g/m^3$ room air) are known to display psychomotor regression (restlessness), auto-aggression behavior (repeated biting of objects or own hands), areflexia (absence of muscle reflex response), ataxia (lack of coordination and balance, such as unable to stand up), erythema, acrodynia, gingivitis (gum disease characterized by irritation, redness and swelling), impaired sensation with hypertension, memory loss and kidney

abnormalities [55-57,60,61,468].

Two cases of such psychomotor and auto aggression behaviors are well-documented: 1) a 11-mo.-old boy exposed to mercury spill from a broken thermometer [469] and 2) a 9-yr.-old boy who dismantled a blood pressure monitor and spilled mercury on his bed and carpeted floor [470]. Their mercury vapor exposures were aggravated by repeated vacuum cleaning. Following medical intervention (DMPS chelation therapy) and proper remediation of mercury contamination of their homes, health of both boys returned to normal in 6 mos.

Also, prolonged exposure to moderate level of mercury vapor $(\leq 50 \ \mu g/m^3 \text{ room air})$ can cause classic mercury poisoning characterized by a triad of signs: tremor, erethism and gingivitis [471]. One such case involved three siblings who were chronically exposed (30-60 d) to mercury vapor in a Grand Rapids, MI home due to a large spill (~20 ml of elemental mercury, resulting in 10-40 μ g/m³ mercury in room air) [472]. They were hospitalized and successfully treated by chelation therapy. In another case of a 10-yr. child in Tehran, Iran who was exposed to mercury vapor at home for 20 days (mercury in blood, 27.7 µg/L and urine, 34.4 µg/L) demonstrated acrodynia, seizure, and visual impairment [473]. The patient recovered after 9 months of treatment with D-penicillamine. The initial brain MRI showed multiple hyperintense lesions in cerebral white matter, left globus pallidus and putamen, and all were resolved with chelation therapy.

WHO [60] lists the following health effects of chronic exposure to various concentrations of mercury in breathing air: $>80 \ \mu g/m^3$ with urine mercury $>100 \ \mu g/g$ creatinine: tremor, erethism and proteinuria; 25-80 $\ \mu g/m^3$ with urine mercury: 30-100 $\ \mu g$ creatinine: tremor, psychomotor disorder, irritability, fatigue and anorexia; 25-35 $\ \mu g/g$ creatinine: tremor and $<30 \ \mu g/g$ creatinine: above listed effects, usually mild and in sensitive individuals only.

ATSDR has set a minimum risk level of $0.2 \ \mu g/m^3$ for mercury in residential air [86]. The following guideline for managing patients exposed to elemental mercury vapor may be used [474]: acute poisoning cases with cough, dyspnea and chest pain, for immediate referral to ER (Grade D) and those with symptoms of chronic toxicity (rash, tremor and weight loss) or exposed to high concentrations of mercury vapor but without dyspnea, for evaluation at a healthcare facility (Grade C).

From exposure to organic mercury compounds

Methyl, ethyl and phenyl derivatives of mercury are of exposure concern to infants and children. While contaminated fish consumption is the major source of MeHg exposure, the use of ethyl and phenyl derivatives as preservative, fungicide and antibacterial agents accounts for their exposures [423]. MeHg is more toxic than the other two organic derivatives of mercury [475].

From low to moderate pre-and postnatal exposure to MeHg

In infants, pre and postnatal (via breastfeeding) exposures to MeHg from maternal consumption of fish are chronic in nature and best assessed by toenail total mercury analysis (safe level $\leq 0.27 \ \mu g/g$) [1].

The major adverse health effects of low-level prenatal MeHg exposure are diminished fetal and infant growth, and neurodevelopmental deficits in language, learning, attention span, vision and motor activities [55,419,476-478]. At moderate level, chronic MeHg exposure may cause: abnormal reflexes, irritability, cognitive deficits, delayed learning, blindness, microcephaly and cerebral palsy [479]. These severe neurotoxic effects occur when maternal hair levels of mercury reach >6 μ g/g (or ~24 μ g/L blood) [480]. The lowest MeHg level in the brain tissue that could cause neuropathological damage is 12 μ g/g [422]. MeHg from fish is a risk factor for age-related cataracts since it readily accumulates in the eye lens [342,421].

From high prenatal exposure to MeHg

Prenatal exposure to high levels of MeHg (maternal blood levels 40-50 μ g/L) during the 1970's mass poisoning in Japan (due to mercury containing industrial waste discharged into river water used for domestic purpose) has caused irreversible damage to fetal CNS resulting in blindness, deafness, cerebral palsy, impaired growth and severe intellectual disability [8,378,479].

ADHD in children caused by maternal MeHg exposure can be treated with the common antidepressant fluoxetine (Prozac) [103,481].

From high postnatal exposure to MeHg

Acute and chronic exposures of children to high levels of MeHg can cause ataxia, dysarthria, visual deficits, hearing loss, peripheral neuropathy, involuntary movements and even death [55,79,80,83,479].

While high-level MeHg exposures are rare, such poisoning on a large-scale has occurred in Iraq in 1971-1972 due to accidental distribution of seed grain treated with a MeHg-based fungicide meant for planting only [55,424,482]. Also, the clinical details of poisoning of a New Mexico family in the US due to the consumption of a hog fed seed grain treated with MeHg fungicide are available [483,484]. Further, extensive poisoning by consumption of fish highly contaminated with MeHg has occurred during the two major industrial disasters in Japan (1953 and 1964-1965). MeHg has a latency period of ~1 mo. after exposure and its lethal dose is 20–60 mg/kg bw with death occurring 2-4 wks after the onset of symptoms [8,80,419].

Information for physicians on recognizing and preventing overexposure [485] and risk assessment [486] of MeHg from fish consumption in children is available.

From acute exposure to ethylmercury (EtHg) in vaccines

EtHg is a component of preservative thimerosal (sodium ethylmercury thiosalicylate) used in children's vaccines. However, it is no longer of exposure concern to children in the US since FDA has banned its use in 1999 [55]. In children, iatrogenic acute exposure to EtHg from vaccines (~25 μ g mercury/vaccine dose equivalent to that in a 3-oz can of tuna) usually leads to blood mercury levels below the EPA safety limit of 5.8 μ g/L. The vaccine-derived mercury in the body is completely eliminated *via* stool by 60 d [308,487].

For Amazonian infants in the ASGM region who are exposed to environmental mercury and MeHg from breast milk and fish diet, additional burden of EtHg from vaccines is relevant only to early neurobehavioral deficits (as measured by GDS scores at 6 mo.) and such deficits are caught up by age 5 yr. [417]. However, the thimerosal-preserved vaccines do increase breastfed infant's hair mercury level [434,488] and they are implicated in Kawasaki's Disease, an acute febrile multiorgan vasculitis that affects children <5 yr. [461].

Extensive studies of children who received thimerosalcontaining vaccines have affirmed their safety [45]. Also, a toxicity review of EtHg used as a preservative in biological (*e.g.*, children's vaccines) and pharmaceutical (*e.g.*, ophthalmic solutions and antibiotic ear suspensions) preparations has attested to its safety [308,489].

From acute exposure to Phenylmercuric Acetate (PMA)

PMA ($C_6H_5HgOCOCH_3$) has been used as an antifungal agent in agriculture, mildew inhibitor in leather products and paints, and additive in eye drops and washes [490]. Also, it is currently used in nasal spray and hemorrhoid relief ointment [308].

Infants exposed to PMA from the fungicide contaminated diapers in Buenos Aires, Argentina have developed acrodynia with increased urinary excretion of γ -glutamyl transpeptidase, a sensitive marker enzyme for mercury toxicity [491]. Absorption of PMA through the skin made infants irritable resulting in profuse sweating, swelling and desquamation (skin peeling) of the extremities, cheeks and nose. Their urinary mercury levels were elevated >50 µg/L which returned to normal along with health after discontinuing contaminated diaper use.

From exposure to inorganic mercury compounds

Inorganic mercury compounds (salts and oxide) are widely used in cosmetics, herbal medicines and dietary supplements. They are water soluble, more reactive than elemental mercury and highly corrosive. Their usual exposure routes are through skin contact and by oral ingestion, the latter known to cause nausea, vomiting and severe abdominal pain [17,34]. Kidney is the primary organ for their accumulation and chronic exposure damages it (indicated by elevated urinary proteins) [34]. Also, the gums become soft and spongy, develop sores and teeth get loose with enhanced saliva flow. In common with other forms of mercury, inorganic compounds also cause neurological damage [492]. For most inorganic compounds, the acute lethal dose is 1-4 g (equivalent to 14-60 mg/kg bw).

There is a correlation between fish consumption and inorganic mercury levels (directly absorbed from fish or formed by demethylation of MeHg in the body) in blood and urine [435]. Also, increase in fetal inorganic mercury levels occurs with increasing number of maternal amalgam fillings [193].

Specific examples of inorganic mercury compounds that children are frequently exposed to are described below.

Mercurous bromide (Hg₂Br₂)

In herbal medical practice, Hg_2Br_2 along with its chloride analog (Hg_2Cl_2) (calomel) are used in popular remedies for bacterial and viral skin infections. FDA has issued strong warning about mercury poisoning of children linked to such usage by adults [493]. Children breathing mercury vapor released from these products when used by adults' exhibit irritability, tremors and other toxic symptoms.

Mercurous chloride (Hg,Cl,) (calomel)

Currently available in the US pharmacies, calomel is used in skin-lightening creams, as purgative, fungicide, and to treat skin infections and itching during chickenpox and other viral/bacterial infections. Formerly used in baby teething powder, calomel can cause mercury poisoning (acrodynia and erythema). Thus, a recent case of a 17-mo.-old toddler's mercury poisoning in an Arizona home was attributed to calomel in skin-lightening cream used by her mother and grandmother [314]. The child exhibited hypertension, fussiness, constipation and arthralgia (temporomandibular joint pain) with high mercury levels in blood (26 μ g/L) and urine (243 μ g/g creatinine). She was successfully treated with DMSA and the contaminated home needed remediation. Also, children in several California families who used such skin-lightening creams were exposed to mercury concentrations >ATSDR safe level, 1.0 μ g/m³ air [315].

Mercuric chloride (HgCl₂)

This mercury salt has been used as a topical anesthetic, to treat ulcers and syphilis; as disinfectant, pesticide and fungicide in agriculture; and as bleach and stabilizer in food industry [17,326]. The classic toxic symptoms of $HgCl_2$ are a combination of renal, GI and CNS impairments that may result in death [494].

Mercuric oxide (HgO)

Ointments containing 1% HgO have been a treatment of choice for *Phthiriasis palpebrarum* infection in the eyelids of children [495]. As described earlier (see Exposure from unregulated cosmetics, etc. section above), FDA allows its use in products intended for eye infections only. The HgO-containing skin whitening and anti-itch creams are currently sold in the US (online and in pharmacies). The use of such products by uninformed parents could put their children in harm's way. Thus, the skin application of a HgO ointment to treat infected eczema has resulted in fatal poisoning of a 4-month-old infant [496]. Death occurred after 32 d and toxic levels of mercury were found in blood, urine, cerebrospinal fluid and vital organs at autopsy.

Since 1996, the use of HgO containing button batteries in toys, watches, smart car keys, etc. is banned in the US [497]. The symptoms associated with ingestion of battery with or without HgO are relatively nonspecific (vomiting, abdominal pain, fever, diarrhea, respiratory distress and dysphagia), and making proper diagnosis can be challenging [498].

Effect of diet and nutrition on mercury poisoning of children

There is wide variation in the neurodevelopmental effects of early-life mercury exposures in children in different parts of the world. This is attributable to the differences in their diet and nutritional status [47,91,416,499].

Several dietary constituents are known to inhibit the adverse effects of mercury [9,86,90,179,227,418,500-503]. They decrease GI absorption and tissue uptake of mercury, and counteract the metal toxicity by their antioxidant activity. For example, tropical fruit consumption reduces mercury uptake in Brazilian children [504,505] and the high intake of vegetables with fish helps reduce MeHg toxicity in Australian children [242,243]. Also, a change in diet from predatory to planktivorous fish is responsible for decreased mercury exposure in reproductive-age women in Brazil due to lower MeHg and higher DHA levels in the latter diet [373]. In contrast, green tea and ethyl alcohol enhance fish MeHg toxicity [86,87].

Similarly, nutrition can play important role in protecting children from toxic effects of mercury. Thus, dietary nutrient Se protects children from mercury toxicity [506]. Conversely, mercury exposure in children with poor nutritional status reduces their response to immunization vaccines, *e.g.*, those in gold mining communities in Peruvian Amazon [507].

Common dietary constituents with modulatory effect on mercury toxicity are described below.

Selenium (Se)

The consumption of Se-rich Ocean fish (yellow fin, skipjack and Blue Marlin) has improved IQ in children by reversing MeHg-induced toxicity in the brain and neuroendocrine system [508]. Dietary Se acts as a scavenger of reactive free-radicals of mercury by forming stable and biologically inert Se-Hg complexes [506,509]. It also acts as a prophylactic or antidotal agent to prevent or reverse mercury toxicity [463,500,510]. Further, Se levels can be used as a bioindicator to monitor mercury exposure and toxicity [511].

Blood Se level increases when freshwater Peacock bass, a predatory species, is consumed with coconut pulp [512]. The Se health benefits can be maximized by preferentially consuming herbivorous (primarily feeding on plants) fish species [513] and Se-rich vegetables, such as cassava root, and Brazil and chest nuts [370,514]. Specifically, the cataractogenic effect of mercury may be offset by Se-rich diet [515].

However, higher amounts of dietary mercury could reduce the protective activity of Se-dependent antioxidant enzymes [516]. Thus, Se: mercury molar ratio >1:1 in fish is required to achieve protection against mercury-induced toxicity [277,517]. This ratio depends on what fish feed on and their geographic location, and it decreases with increase in fish size and age [291,518-521]. Further, larger fish (>90 cm) exhibit biomagnification of Se and mercury, while smaller ones (<80 cm) do not [522]. Farming is currently being utilized to improve quality and safety of fish by reducing MeHg level and enhancing Se content [51].

While maternal iron status in blood appears to be unrelated to MeHg-induced neurotoxicity [375], combined high levels of lead (>31.4 μ g/L) and mercury (>12.7 μ g/L) have been associated with chronic under-arousal psychopathology in adolescent boys [523].

Tropical fruits and other constituents rich in antioxidants

Recent epidemiological studies in Brazil found that consumption of common tropical fruits (*e.g.*, bananas and oranges) can reduce both short-and long-term mercury uptakes in the body (as reflected by ~30% decreases in blood and scalp hair mercury levels) [504,505]. Also, wheat barn reduces mercury concentration in the brain [83]. MeHg toxicity is counteracted by its binding with dietary: 1) antioxidants, *e.g.*, DHA in fish and eggs, ascorbic acid (Vitamin C) in fruits and vegetables, phenolic gingerols and shogaols in ginger, polyphenols in coffee, tannic acid and flavonoids in tea, and 2) SH group containing glutathione in fruits and S-allyl cysteine compounds in garlic, ginger and turmeric. Such MeHg binding interferes with its absorption and enhances excretion from the body [48,105,504,524-526]. Thus, consuming broiled or fried fish with tea or coffee can lead to decreased bioaccessibility and enhanced excretion of MeHg contaminant [279,502]. The dietary DHA provided by egg yolk is more economical and convenient to consume compared to fish [105].

Annatto, the seed of South American shrub *Bixa orellana* L. is used for commercial production of butter, margarine, cheese and ice cream. It contains antioxidant carotenoid pigments with potential protective effect against MeHg toxicity [527].

Ethyl alcohol (alcohol)

In EU, children start drinking wine at age 12 to 14 yr. and in many other counties it is legal for 16-yr-old to drink alcoholic beverages. Alcohol consumption is known to decrease body retention of inhaled mercury vapor by enhancing its elimination in exhaled air [528,529]. This result in mercury: 1) storage reduced in lungs, 2) level lowered in blood and 3) storage increased in liver.

A study of US dentists has confirmed the inhibitory effect of alcoholic drinks on the body uptake of occupationally exposed mercury vapor (while working with dental amalgam) based on its decreased excretion in the urine [530]. Alcohol is an inhibitor of catalase enzyme which is essential for oxidizing elemental mercury before it is absorbed in the body. In contrast, alcohol consumption increases MeHg toxicity, especially in the kidney [83,87].

Processed foods

Blood inorganic mercury, attributable to processed food intake, is directly associated with lower blood glucose levels [531]. This can result in reduced risk factors for type 2 diabetes. Children are routinely exposed to trace amounts of HgCl_2 , (used in food industry for bleaching flour, refining vegetable oil products, and as inhibitor of corn starch degradation) when they consume processed foods, especially those containing high fructose corn syrup [326]. The latter, when used with certain artificial food colors (*e.g.*, sunset yellow) can lead to loss of zinc which is essential for elimination of mercury from the body [90].

Additionally, high fat diet, pasteurized milk, white bread, French fries, cornstarch, wheat bran and flour have shown protective effect against mercury toxicity by enhancing its elimination from the body [82,313].

Besides dietary constituents and nutritional status, factors such as gender, genetics, pharmacodynamic variables, maternal smoking, co-exposure to other pollutants and local contexts along with socio-environmental variables may also affect individual child's vulnerability and response to similar mercury exposures [10,96,159,160,177,178,418,487,502].

Prevention of mercury poisoning of children

The pediatric exposure to mercury can be minimized by salubrious parents with the knowledge of its exposure sources

and resources available to take preventive measures outlined in this review.

Also, pediatricians, family practitioners and other medical professionals involved in emergency and poison control management could play important role in educating parents on the exposure hazards of mercury. Further, school science teachers should be active partners in educating children on the toxicity and safe handling of mercury.

To assist the above groups, this review has provided latest information on the following topics:

- 1. Common sources of mercury exposure at home (broken fever thermometers, OTC and herbal medicines, dietary supplements, cosmetics, tainted toys and jewelry).
- 2. How to handle small mercury spills at home, such as from a broken fever thermometer.
- 3. Major mercury spills and other contamination issues at home.
- 4. Why parents should not extract gold from scrap materials at home that involves heating mercury.
- 5. Hazards of mercury used in religious and cultural ceremonies at home.
- 6. Science teachers in schools should store mercury securely under lock and maintain proper inventory.
- 7. Educating children in school that mercury is not a toy but a hazardous metal.
- 8. In school science laboratories, students should handle mercury only under the supervision of a teacher. Facemask and gloves are needed for safe handling of mercury.
- 9. Pediatricians and other medical professionals should familiarize themselves with sources of mercury contamination, signs and symptoms of poisoning by the different forms of mercury.
- 10. Which fish to consume that has low mercury content?
- 11. The role of dietary constituents, such as tropical fruits in counteracting mercury toxicity from contaminated fish consumption.
- 12. Check ingredient label for mercury before purchasing any product suspected of contamination.

In some schools in the US, instead of teaching students safe handling of mercury, it is being banned in science laboratories [532]. Since human crematoriums account for significant atmospheric mercury pollution (several mg/m³ per cremation; India alone emits 1.4 tons/ year) [13,352], extraction of amalgam filled teeth of the diseased prior to cremation is worth considering.

Conclusions

From Fresno, California in the US to Denizli, Turkey and Shanghai, China, mercury poisoning of children continues to be of public health concern around the world with ongoing reports of nervous system damage, critical organ toxicity, pulmonary edema and even death. Because of its ubiquitous nature, mercury is currently considered as the third toxic substance in the global environment and its exposure in children is expected to continue indefinitely.

However, there is indication that such environmental mercury exposure of children is being reduced by the development of mercury-free consumer devices (*e.g.*, digital fever thermometer and LED light bulb) and materials (*e.g.*, synthetic dental fillings without mercury), alternate manufacturing and non-fossil power generating processes, and improved occupational safety measures. With the continued progress in these fronts, the prospects for curtailing environmental mercury poisoning of children around the globe are bright indeed. There is high hope for the latest discovery of bioremediation of MeHg-polluted soil and water by the common plant symbiotic fungus *Metarhizium robertsii* to provide a viable solution for the global environmental mercury pollution [533,534]. The fungus enzymes, MeHg demethylase and mercury ion reductase detoxify MeHg by breaking it down to less toxic metallic mercury (Hg⁰).

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