# Comparative study on the chemical composition of *Corchorus olitoriusl* leaf and stem dry oils.

# Hanan M. Al-Yousef<sup>1\*</sup>, Musarat Amina<sup>1</sup>, Syed Rizwan Ahamad<sup>2</sup>

<sup>1</sup>Department of Pharmacognosy, College of Pharmacy, King Saud University, Riyadh, Saudi Arabia

<sup>2</sup>Central Laboratory, Department of Pharmaceutical Chemistry, College of Pharmacy, King Saud University, Riyadh, Saudi Arabia

#### Abstract

A comparative study was done on the dry oils isolated from the leaves and stem portions of *Corchrus olitorius* (Linn.) growing in Saudi Arabia. Oils were obtained by column chromatography and analysed by Gas Chromatography (GC), Gas Chromatography/Mass Spectrometry (GC/MS) and <sup>13</sup>C NMR spectroscopy. The investigation led to the identification of 22 components in leaves oil while as 23 components in stem oil of *C. olitorius*. On comparison of leaves and stem dry oil components, both the oils were rich in hydrocarbons and fatty acids, but the leaves were found to contain higher percentage of hexadecanoic acid (7, 28.52%) and 2, 4-di tert-butyl phenol (2, 15.01%) as a main components whereas stem dry oil contains ethyl palmitate (6, 26.19%) and 2, 4-di tert-butyl phenol (1, 14.35%) in higher percentage respectively. Also dry oils were tested for antimicrobial and antioxidant activities. Moderate antimicrobial activity with MIC-values between 0.40-0.8 and 1.6->3.2 mg/ml for components of leaves and components of stem oil respectively were observed while as significant anti-oxidant effects were exhibited by both the oil components for DPPH-radical scavenging assay at a dose dependent manner of *C. olitorius* leaves oil (95.1%) and stem oil (97.1%) at 400 µg/ml respectively.

Keywords: Corchorus olitorius L, GC/MS, <sup>13</sup>C NMR, Antimicrobial activity, Antioxidant activity.

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### Introduction

Corchorus olitorius (Linn.) belong to the family Tiliaceae and commonly known as "Jute or Jew Mallow". It is disturbed throughout the tropical region of Africa and Asia [1] and is considered as a common vegetable in Egypt, the Philippines, Australia, Senegal and Thailand [2], as a common weed for India, Afghanistan, Nepal, Kenya and Turkey and as a principle weed for Sudan [3]. Its tender leaves and shoots, rich in proteins and minerals servers as a main dietary source for proteins and are used in ethnic soup preparation in several tropical countries [4]. In the literature survey, Corchorus olitorius is ascribed for several medicinal uses [5]. Its leaves have been reported to be used for the treatment of pain, gonorrhoea, chronic cystitis and tumors [6]. The dried young leaves are diuretic, demulcent, tonic and slightly febrifuge; infusion of dried leaves is used to restore appetite and strength [7]. Its seeds are purgative and have been reported to exhibit estrogenic activity [8] as well as effective in cardiac diseases due to presence of appreciable amounts of active cardiac principles in particular Olitoriside, which showed equivalent effect to strophanthin with chronic cardiac patients [9]. The stem acts as a main source of jute fiber [10]. In addition to this, it is an ingredient in lotions, facial and hand creams and hair tonics [11]. Moreover, pharmacologically C. olitorius

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possesses a diverse biological activities which includes, antioxidant [12], anti-tumor [13], hypoglycemic [14], antimicrobial [15], anti-inflammatory, analgesic [7], antiobesity [16], gastroprotective [17] and wound healing effects [18]. Further more extensive phytochemical investigation showed presences of cardiac glycosides: coroloside, veticoside, erysimoside, helveticoside, corchoroside A, corchoroside, B, strophanthidol, evonoside, strophanthidin, olitor and chorchorusoside A-E from the seeds [19]; triterpenes, intones, steroids, acidic polysaccharide rich in uronic acid and consisting of rhamnose, glucose, galacturonic acid and glucuronic from the leaves [20] while as root contains corosin,  $\beta$ -sitosterol and triterpene [21].

Thus the excellent chemical and biological profile of *C*. *olitorius* promoted to further investigate the useful chemical entities from different parts of the plant, which led to isolation of dry oils from the leaves and stem portion of *C*. *olitorius*. The *C*. *olitorius* was collected from local market of Saudi Arabia, where it is locally known as 'Molokhia' and consumed as a popular vegetable. The present paper describes isolation of dry oils, identification of components and comparison leaves and stems dry oil by using GC/Gas Chromatography linked to Mass Spectroscopy (GC-MS), Carbon-13 nuclear magnetic resonance ( $^{13}C$  NMR) technique and evaluation of

antimicrobial and antioxidant activities of each dry oil obtained. To the best of our knowledge this the first comparative study of dry oils obtained from different portions of C. *olitorius*.

### **Materials and Methods**

### Plant material

The fresh plant was purchase from local market at Riyadh, S.A in June 2016. The leaves and stem of *C. olitorious* were separated from each other, air-dried and then coarsely powdered to give powdered leaves (456 g) and stem (65 g). The plant materials were identified by Dr. Mahmoud Abdul Aziz Mahmoud, College of Food and Agricultural Sciences, KSU. Voucher specimens (HM37-1) for leaves and (HM37-2) for stem were deposited in department of Pharmacognosy, college of pharmacy, KSU.

### **Solvents**

The solvents used were ethanol 95%, acetone, petroleum ether 40-60°C, chloroform, methanol, butanol and ethyl acetate, which were distilled prior to use. Analytical grade solvents were used for chromatography and crystallization while those used for extraction processes were General Purpose Reagents (GPR). Spectroscopic grade and deuterated solvents were used for spectral analyses.

### Extraction and isolation

The leaves and stem of C. olitorius were separated from each other, air dried and then powdered leaves (456 g) and stem (65 g) were exhaustively extracted with 96% alcohol at room temperature. The alcoholic extract was evaporated under vacuum to obtain 45.1 g and 63.3 g of dark green residues for leaves and stem respectively. Each residue obtained was suspended in mixture of water-methanol (9:1) individually and successively partitioned successively with petroleum ether (3  $\times$ 0.5 L), chloroform (3  $\times$  0.5 L), ethyl acetate (3  $\times$  0.5 L), nbutanol  $(3 \times 0.5 \text{ L})$  to get petroleum ether (CCL1, 4.8 g), CHCl<sub>3</sub> (CCL<sub>2</sub>, 5.2 g), EtOAC (CCL<sub>3</sub>, 4.2 mg), n-BuOH  $(CCL_4, 5.0 \text{ g})$  for leaves and petroleum ether  $(CCS_1, 5.4 \text{ g})$ , CHCl<sub>3</sub> (CCS<sub>2</sub>, 6.2 g), EtOAc (CCS<sub>3</sub>, 9.5 mg), n-BuOH (CCS<sub>4</sub>, 8.5 g) for stem residue respectively. All the fractions were stored in refrigerator prior to its use. The chloroform fraction for leave (CCL2, 5.2 g) and stem CHCl<sub>3</sub> (CCS2, 6.2 g) were individually chromatographed over silica gel column (250 g, 2.5 cm) and eluted with n-hexane-ethylacetate (8:2), gradually increasing the percentage of ethyl acetate in n-hexane, nhexane-ethylacetate (7:3), n-hexane-ethylacetate (6:4) and final elution with 100% ethyl acetate. Fractions eluted in n-hexaneethylacetate (7:3) were combined on the basis of similar TLC pattern and subjected to recrystallization to obtain 8.2 mg of greasy solid for leaves and 12.0 mg for stem respectively.

#### Gas chromatography-mass spectroscopy analysis

The GC-MS analysis was performed in a Perkin Elmer Clarus 600 gas chromatograph inked to a mass spectrometer (Turbomass) available at Central Laboratory, College of Pharmacy, King Saud University, Riyadh. An aliquot of 2  $\mu$ L of extract was injected into the Elite-5 MS column of 30 m, 0.25  $\mu$ m film thickness, 0.25  $\mu$ m internal diameters.

# Capillary column using the following temperature program

The GC-MS system starts with the initial oven temperature of 40°C and hold for 2 min then up to 200°C increase with the rate of 5°C/min and hold to 2 min. From 200°C again at the rate of 5°C reach the column oven temperature to 300°C and hold for 5 min. The injector temperature was maintained at 280°C. The interface temperature was 240°C and the source temperature was at 220°C. Pressure of the system was maintained at 1.11 e<sup>-5</sup> and the electron energy was at 70 eV. Helium was used as a mobile phase at a flow rate of 1.0 ml/min. Mass spectral detection was carried out in electron ionization mode by scanning at 40 to 600 (m/z). Finally, unknown compounds were identified by comparing the spectra with that of the National Institute of Standard and Technology and WILEY library. The total time required for analysing a single sample was 61 minutes.

# <sup>13</sup>C-NMR

<sup>13</sup>C-NMR spectra of samples components of leaves and components of stem were recorded on a Bruker AM700 operating at 176.0 MHz for <sup>13</sup>C NMR, (Bruker Topspin Gmbh, Rheinstetten) NMR spectrometer (Research center, college of pharmacy, King Saud University) equipped with a 5 mm probe, in deuterated chloroform. The spectra were recorded with the following parameters: pulse width (PW): 12 µsec, Acquisition Time (AQ): 0.786432 sec for 295.0 K data table with Spectral Width (SW) of 41666.668 Hz (250 ppm). The number of accumulated scans was (NS) 3000 for each sample. An exponential multiplication of the free induction decay with the line broadening of 0.1 Hz was applied before transformation.

### Identification of components by GC-MS

The components were identified on the basis of GC retention time and matching with Wiley 2006 library as well as by comparison the fragmentation patterns of their mass spectra with those reported in the literatures [22,23] and most of the identified components were higher alkanes and alcohols. A total of 22 and 23 detectable peaks were selected for Components of leaves oil and components of stem oil respectively.

### **Biological evaluation**

Antimicrobial activity: Microorganisms: American Type of Culture Collection (ATCC) standard against various microorganisms namely, *Staphylococcus aureus* (ATCC25922), *Escherichia coli* (ATCC25923), *Pseudomonas aeruginosa* (ATCCPAO1) and *Candida albicans* (SC315) were used.

#### Antimicrobial assay

An earlier adopted agar well diffusion method, [24,25] was used. The 0.1 ml of diluted inoculum ( $10^5$  CFU/ml) of test organism was spread on Muller-Hinton agar plates. Wells of 8 mm diameter were punched into the agar medium and filled with 100 µl of plant extracts and 100 mg/ml concentration of dry oils and solvent blank (DMSO) separately. The plates were incubated for overnight at 37°C. The antibacterial activity was evaluated by measuring the zone of inhibition against test organism. Antibiotics (Ampicillin and Doxycycline) were used as positive control against bacteria while Nystatin was used as the control antifungal drugs. Each experiment was performed in triplicate [26].

# Determination of minimum inhibitory concentration (MIC) of plant extracts and dry oils

Minimum inhibitory concentration of plant extracts as well as dry oils against drug resistant clinical strains was determined by broth dilution method, using specific dye (p-iodonitro tetrazolium violet) as an indicator of growth as described by [27]. Briefly, 2 ml of the plant extracts and dry oils were mixed with 2 ml of Muller-Hinton broth (Hi-Media Ltd., Mumbai, India) and serially diluted into the next tube and so on. 2 ml of an actively growing culture of different test strains was added before incubating for overnight, at 37°C. After examining turbidity visually, 0.8 ml of 0.02 mg/ml indicator dye (piodonitro tetrazolium violet) was added to each tube and incubated at 37°C. The tubes were examined for the color development, after 30 min. Absence of growth was also confirmed by spreading 0.1 ml of broth from each test tube on normal nutrient agar plates. MIC is defined as the minimum concentration of plant extracts as well as dry oils which inhibited the visible growth of test strains.

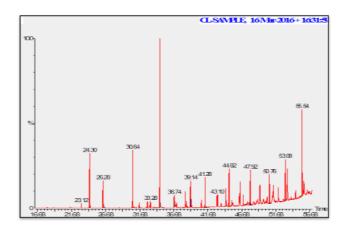
#### Antioxidant assay

DPPH radical scavenging assay: Free radical scavenging activity of different plant extracts and dry oils against stable 1,1-diphenyl-2-picrylhydrazyl (DPPH) was determined spectrophotometrically by slightly modified method of Gyamfi et al. [28] as described below. When DPPH reacts with an antioxidant, which can donate hydrogen, it is reduced. The changes in color (from deep-violet to light-yellow) were measured at 517 nm on a UV/visible light spectrophotometer (Spectronic 20 D+, Thermo Scientific, USA). Fifty µl of the solvent dried extracts in methanol, yielding different concentrations was mixed with 1 ml of 0.1 mM DPPH in methanol solution and 450 µl of 50 mM Tris-HCl buffer (pH 7.4). Methanol (50 µl) was used as a vehicle control in the experiment. After 30 min of incubation at room temperature the reduction of the DPPH free radical was measured spectrophotometrically. Ascorbic acid was used as positive controls. Inhibition percent was calculated from the following equation:

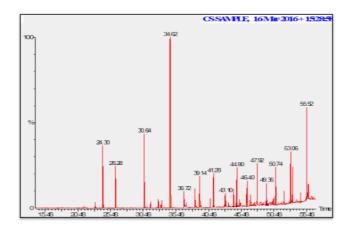
Inhibition (%)=(Absorbance of control-Absorbance of test sample/Absorbance of control)  $\times$  100

#### Statistical analysis

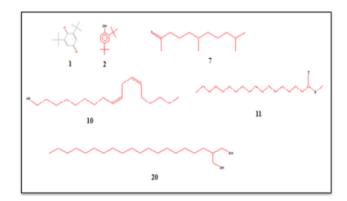
All experiments were performed in triplicates and the data obtained from experiments were presented as mean values and the difference between control and test were analysed using student's t test. Statistical analysis is done on Sigma plot 12.



*Figure 1. GC/MS* chromatogram of the dry oil of *C.* olitorious leaves *(CL)* growing in SA.



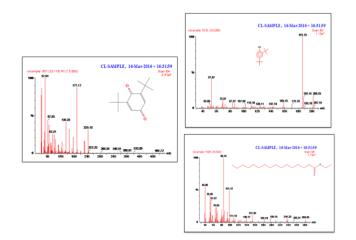
*Figure 2. GC/MS* chromatogram of the dry oil of C. olitorious stem (CS) growing in SA.



*Figure 3.* Major compounds isolated from C. olitorious leave (CL) growing in Saudi Arabia.

#### **Results and Discussion**

C. olitorius is a native plant of tropical Africa and Asia. Its leafy vegetable is popularly used in soup preparation and folk medicine. A comparative study was carried out on dry oils isolated from the leaves and stem parts of C. olitorius. Pale vellow coloured dry oil with a distinctive odor, was obtained by individual column chromatography of chloroform fractions of leaves and stem portion of C. olitorius. Isolated dry oils were analysed by Gas Chromatography (GC), Gas Chromatography/Mass Spectrometry (GC/MS) and identification of individual components was done by <sup>13</sup>C NMR spectroscopy. Gas chromatography and Gas chromatography-Mass spectroscopy analysis of isolated dry oils showed the presence of twenty two components in leaves oil (Table 1) while as twenty three components in the stem oil (Table 2) respectively, which were identified by <sup>13</sup>C NMR. On comparison between leaves and stem dry oil, results showed that both the oils are composed of similar types of hydrocarbons and fatty acids, but the percentage of hexadecanoic acid (7, 28.52%), 2, 4-di tert-butyl phenol (2, 15.01%), tetratetracontane (19, 7.77%), tritetracontane (22, 6.17%) and 9-eicosene (4, 5.31%) components was in leaves oil whereas stem oil contains ethyl palmitate (6, 26.19%), 2, 4di tert-butyl phenol (1, 14.35%), 1-eicosanol (23, 8.27%), tetratetracontane (16, 7.02%), 8-heptadecene (3, 5.33%), hexatriacontane (18, 5.16%) in higher percentage (Figures 1-6). The <sup>13</sup>C NMR spectra were recorded in CDCl<sub>3</sub> showed approximately similar protonated carbons signals in both components of leaves oil as well as stem oil (Figures 7 and 8), which is consistent with the GC and GC/MS results and with those common hydrocarbons. Thus the results obtained describe the first detailed analysis of C. olitorious leaves and stem dry oil. Our results showed slightest differences between the components of leaves and stem dry oils and also describes the variations of the contents of the main components, the plant produces dry oil belonging to the same chemotype during its life.



*Figure 4.* A mass spectrum of some dry oil components of C. olitorious leaves (CL) growing in Saudi Arabia.

#### **Biological evaluation**

Dry oils were evaluated for their antimicrobial and antioxidant activities. Results showed that both dry oils from C. olitorious possessed antibacterial and antifungal activity against all of the tested gram positive and gram negative bacterial and fungal strain (Tables 3 and 4), with the diameters of zone inhibition ranging between 11-13 mm and 13-16 mm for components of leaves oil and components of stem oil respectively. The most significant activity was observed for components of stem dry oil and inhibited the growth of all the bacterial strains tested, specifically S. aureus, E. coli (16 mm). Furthermore, antifungal results showed, C. albicans was susceptible to components of leaves oil and components of stem dry oils (11 and 16 mm), with MIC-values between 0.40-0.8 and 1.6->3.2 mg/ml for components of leaves oil and components of stem dry oil respectively. Also study on was conducted petroleum ether extract of C. olitorius leaves and offered a good activity against Escherichia coli, Staphylococcus aureus and Yersinia enterocolitica, 20 mm, 19 mm and 19 mm, respectively. The results concluded in this study appear to emphasize on the antibacterial and antifungal potential of C. olitorius leaves, as well as its helpfulness in the treatment of diseases that might be as a cause of infection [29]. Moreover, the DPPH-radical scavenging assay exhibited a significant antioxidant activity at a dose dependent manner of C. olitorius components of leaves oil and components of stem dry oils (95.1 and 97.1%) at 400 µg/ml (Table 5) respectively.

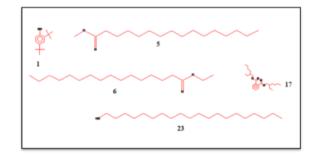
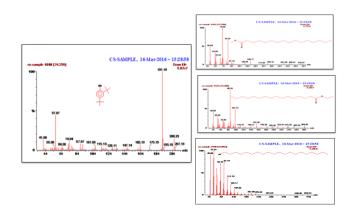


Figure 5. Major compounds isolated from C. olitorious stem (CS) growing in Saudi Arabia.



*Figure 6.* Mass spectra of some dry oil components of C. olitorious stem (CS) growing in Saudi Arabia.

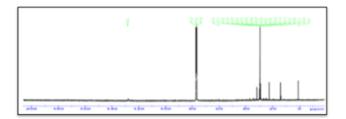
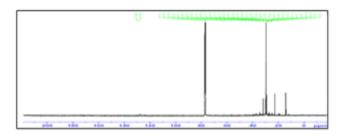


Figure 7. <sup>13</sup>C NMR spectrum (176.0 MHz,  $CDCl_3$ ) of C. olitorious leaves (8.2 mg, CL). NS=3000, AQ=0.786432 sec.



*Figure 8.* <sup>13</sup>*C NMR spectrum (176.0 MHz, CDCl<sub>3</sub>) of C. olitorious stem (12 mg, CS). NS=3000, AQ=0.786432 sec.* 

**Table 1.** Chemical composition of the dry oil of the C. olitorious leaves (CL) growing in SA.

S.no	Name	R	Area	M+
1	2, 6-di-butyl-2, 5-cyclohexadiene	23.14	0.52	220.18
2	2,4-Di tert-butyl phenol	24.3	15.01	206.23
3	1-Heptadecanol	26.28	2.67	256.46
4	9-Eicosene	30.64	5.31	280.2
5	2-Undecanone	31.62	1.09	198.34
6	Heptadecanoic acid	33.28	1.44	270.27
7	Hexadecanoic Acid	34.62	28.52	255.31
8	Isoheptadecanol	36.74	1.73	256.46
9	2-propyldecan-1-ol	37.1	0.96	200.31
10	9, 12-octadecadien-1-ol	38.38	1.58	266.4
11	Octadecanoic acid	39.14	3.85	312.2
12	Pentadecane	39.28	0.81	212.06
13	9-eicosene	40.74	0.77	280.2
14	Tetratetracontane	41.28	5.89	619.2
15	Heptadecanoic acid	43	1.06	270.1

16	Hexatriacontane	43.1	3.87	506.97
17	Octadecanal	43.64	1.03	268.49
18	10-methyl-1-hexadecanol	44.36	1.44	256.46
19	Tetratetracontane	44.82	7.77	619.2
20	2-octadecyl-propane-1,3-diol	46.92	1.12	328.57
21	Tetratetracontane	47.92	7.36	619.2
22	Tritetracontane	50.76	6.17	605.15

**Table 2.** Chemical compositions of the dry oil of the C. olitorious stem(CS) growing in SA.

S.no	Name	RT	Area %	M+
1	2, 4-di tertiary butyl-phenol	24.3	14.35	206.23
2	1-heptadecanol	26.28	3.03	256.46
3	8-heptadecene	30.64	5.33	238.45
4	2-pentadecanone	31.6	1.02	226.39
5	Methyl palmitate	33.28	1.34	260.27
6	Ethyl palmitate	34.6	26.19	284.28
7	1-octadecanol	36.72	2.37	270.49
8	Eicosane	37.1	0.85	282.56
9	(Z, Z)-heptadeca-8, 11-dien-1-yl	38.36	1.57	402.56
10	Heptadecanoic acid	39.12	3.33	270.27
11	5-eicosene	40.72	0.68	280.53
12	Tetratetracontane	41.28	5.14	619.2
13	Heptadecanoic acid	43	0.87	270.1
14	1-octadecanol	43.6	0.44	270.49
15	5-eicosene	44.36	1.56	280.53
16	Tetratetracontane	44.8	7.02	619.2
17	1, 2-benzenedicarboxylic acid	45.24	1.31	390.55
18	Hexatriacontane	46.4	5.16	506.97
19	Hexadecanal	46.92	1.09	240.42
20	10-methyl-1-hexadecanol	50.42	0.65	256.46
21	Tetratetracontane	50.74	6	619.2
22	Tritetracontane	52.08	2.46	605.15
23	1-eicosanol	55.52	8.27	298.55

Table 3. Antimicrobial activity of CL and CS dry oils as inhibition zones (mm).

Dry oils	Zone of inhibition (mm)			
	S. aureus	E. coli	P. aeruginosa	C. albicans
CL	13 ± 0.4 (61.9%)	11 ± 0.3 (44.0%)	15 ± 1 (62.5%)	11 ± 0.6 (47.8%)
CS	16 ± 2.5 (76.2%)	16 ± 1.6 (64.0%)	13 ± 0.5 (54.2%)	16 ± 0.8 (69.6%)

0.8

0.8

Ampicillin	21	-	-	-	
Doxycycline	-	25	24	-	
Nystatin	-	-	-	23	

CL

0.4

**Table 4.** Minimum Inhibitory Concentration (MIC) of CL and CS dry oils against bacterial and fungal strains.

Dry oils	Minimum inhibitory concentration (mg/ml)			
	S. aureus	E. coli	P. aeruginosa	C. albicans

CS	1.6	>3.2	>3.2	
*Values	of MIC are giv	ven in v/v (%) fo	r dry oils	

0.4

0.8

Table 5. Free radical scavengi	ng activity of dry oils by DPPH.
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Dry oils	Percent decolorization	on by DPPH method				
	Concentrations of dr	y oils (µg/ml)				
	12.5	25	50	100	200	400
CL	4.64 ± 2.37 (13.3%)	42.54 ± 3.87 (61.2%)	55.85 ± 2.07 (63.9%)	72.04 ± 3.70 (78.7%)	82.53 ± 2.56 (87.2%)	90.53 ± 4.06 (95.1%)
CS	6.23 ± 1.16 (17.8%)	43.41 ± 2.7 (62.4%)	59.77 ± 1.77 (68.4%)	77.97 ± 2.3 (85.2%)	84.53 ± 3.71 (89.3%)	92.46 ± 2.94 (97.1%)
Ascorbic acid	34.91 ± 1.98	69.56 ± 2.64	87.41 ± 1.47	91.5 ± 2.16	94.66 ± 2.18	95.20 ± 1.65

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# **Conflict of Interest**

None

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# \*Correspondence to

Hanan M. Al-Yousef

- Department of Pharmacognosy
- College of Pharmacy
- King Saud University
- Saudi Arabia