

Plant synthesis of silver nanoparticles using *Matricaria chamomilla* plant and evaluation of its antibacterial and antifungal effects.

Masoud Negahdary¹, Shahin Omid², Ali Eghbali-Zarch¹, Seyed Abdolhossein Mousavi³, Gholamreza Mohseni⁴, Yazdan Moradpour⁵, Ghasem Rahimi^{6*}

¹Yazd Cardiovascular Research Center, Shahid Sadoughi University of Medical Sciences, Yazd, Iran

²Department of Biology, Islamic Azad University, Arsanjan Branch, Arsanjan, Iran

³Department of Health Services Management, Bushehr University of Medical Sciences, Bushehr, Iran

⁴Department of Anesthetics, Shahid Beheshti University of Medical Sciences, Tehran, Iran

⁵Department of Plant Pathology, Marvdasht branch, Islamic Azad University, Marvdasht, Iran

⁶Young Researchers and Elite Club, Marvdasht Branch, Islamic Azad University, Marvdasht, Iran

Abstract

One of the significant goals in nanotechnology is producing nanoparticles with the ability to control and reduce costs and environmental pollutions. Recently, production of nanoparticles using living biological systems has resulted in creation of a new branch of nanotechnology. Silver nanoparticles are among the most widely used nanoparticles. This study aimed to produce silver nanoparticles using *Matricaria chamomilla* and to evaluate its antibacterial and antifungal effects. First *Matricaria chamomilla* extract was prepared by boiling. Then, reaction occurred in the dark and in room temperature by placing the extract in the vicinity of 20mM silver nitrate. Physicochemical properties of the produced nanoparticles were explored using a spectrophotometer device and an electron microscope. Eventually, the antibacterial and antifungal properties of these nanoparticles were evaluated. After the study of the spectrum of UV-Vis and electron microscope it was found out that *Matricaria chamomilla* is capable of producing silver nanoparticles in extracellular form with range of 60 to 65 nm and with different morphologies. It was also found out that these nanoparticles have good antibacterial and antifungal properties against the bacteria and yeast that were used in this study. *Matricaria chamomilla* is a reducing agent that is suitable and compatible with environment for producing nanoparticles.

Keywords: Plant synthesis of nanoparticles, *Matricaria chamomilla*, silver nanoparticles, antibacterial effects, antifungal effects

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Introduction

Nanotechnology is an interdisciplinary field and is formed from the convergence of chemistry, physics and biology. We see unique physicochemical properties when the size of the particles is reduced. In fact, with entering the realm of nanotechnology, we see that surface area to volume ratio and quantum behavior are the most important characteristics that are seen during the shrinking of the size of particles [1]. Because of the low toxicity to cells in the body and high surface plasmon resonance, silver nanoparticles have numerous applications in diagnosis and treatment of diseases, particularly cancer, use in catalysts and biosensors and elimination of bacterial and fungal infectious agents[2]. Achieving nanoparticles with the ability to control size and shape and synthesis of nanoparticles

that have desirable properties and are compatible with the environment are among the goals of nanotechnology. So far, different methods have been defined for the synthesis of nanoparticles in the field of nanotechnology. Each of these methods has advantages and disadvantages depending on the reactions used. Currently nanoparticles are commercially synthesized by chemical method. These methods have resulted in the creation of a production attitude that is appropriate and compatible with the environment due to environmental pollutions resulted from the material used in chemical synthesis such as reducing agents for reducing metal ions and stabilizing agents for preventing the accumulation of nanoparticles that are used in various processes, low yields due to excessive consumption of energy to produce nanoparticles, high cost of production and the low quality of nanoparticles produced

due to high temperatures [3, 4]. Recently, biosynthesis of nanoparticles using biological systems such as bacteria, fungi, yeasts and plants has been defined as a new branch of nanotechnology due to the high effectiveness and efficiency and being compatible with the environment [5,6]. Nanoparticle production using plants can be useful for different purposes considering the ability to control the size and dispersion level of particles. The production of nanoparticles using plant extracts has been proved by many researchers [7]. This study was conducted with the purpose of plant synthesis of silver nanoparticles using *Matricaria chamomilla* and assessment of its antibacterial and anti-fungal effects.

Materials and Methods

All materials and equipment used in this study including sabouraud dextrose agar media, sabouraud dextrose broth media, Nutrient broth media, Mueller- Hinton agar media, and silver nitrate for the synthesis of nanoparticles were purchased from Merck company in Germany. Standard strain of the yeast *candida albicans* (ATCC 14053) and *staphylococcus aureus* (ATCC 25923) were purchased from the Center of Scientific and Industrial Research of Iran.

Production of Silver nanoparticles from Matricaria chamomilla

In this research silver nanoparticles were synthesized from *Matricaria chamomilla* extract. Boiling method was used to prepare extract of dried *Matricaria chamomilla*. First, 25 gr of dried *Matricaria chamomilla* was placed in an Erlenmeyer flask and its volume was brought to 100 ml using deionized water. Erlenmeyer flask contents were boiled for 5 minutes on flame. The extract obtained using this method was filtered using filter paper and 0.45 μm microfilters after cooling. Then 20 mM silver nitrate was added to the extract. The lid of the container containing extract and silver nitrate was closed and the container was kept in a dark place for reaction [8].

Exploring physic-chemical characteristics of nanoparticles

UV-visible spectroscopy

In order to explore the UV-visible characteristics of silver nanoparticles synthesized from *Matricaria chamomilla* 200 μm of the solution resulted from reaction of the extract of the aforementioned plant and silver nitrate was brought to the volume of 1 ml in cuvette and the spectroscopy was done using spectrophotometer device (at wavelengths of 400-800 nm). Silver nitrate solution with *Matricaria chamomilla* extract at zero time was used as the control group [8].

Electron microscopic study of silver nanoparticles

For electron microscopic study of nanoparticles produced from *Matricaria chamomilla*, sediment resulted from the reaction of the plant extract and silver nitrate was centrifuged for 5 minutes using a centrifuge device at the speed of 12000 rpm. The sediment resulted in this stage was used to investigate the electron microscope [8].

Exploring the antibacterial effect of the produced nanoparticles

For exploring the antibacterial effects of the produced nanoparticles the effects of these nanoparticles was explored on *staphylococcus aureus* ATCC (25923) in NB media culture. To evaluate the effect of nanoparticles in liquid media, 100 μl of a suspension of nanoparticles with concentration of 0.37×10^8 Particles ml^{-1} in NB medium containing 1×10^7 cellml^{-1} bacterial suspension was inoculated. The tube containing suspension of silver nanoparticles and bacteria was incubated at 37 ° C for 48 h. To assess the effects of these nanoparticles, the optical density (OD) was measured in spectrophotometer at wavelength of 600 nm and 0, 2, 4, 6, 8, 10, 12, 24, 48 times and its diagram was drawn. The media containing bacterial suspension without any nanoparticles were used as control group [9, 10].

Exploring the antifungal effect of the produced nanoparticles

To evaluate the antifungal effects of nanoparticles synthesized from *Matricaria chamomilla* the effect of nanoparticles on biofilm formation in pathogenic yeast *Candida albicans* (ATCC 14053) was evaluated. The biofilm of this yeast was formed in 99-well microplate. After pouring 100 μl of fungal suspension containing 1×10^6 cell in each milliliter into wells of microplate, the microplate was placed in a shaking incubator at 37 ° C with 75 rpm in order for the cells to attach to the bottom of the well. To isolate unattached from buffer, phosphate buffered saline with pH 7.4 was used in a way that each well was washed three times with 100 μl of phosphate buffered saline with pH 7.4. Then 100 μl of medium (SDB) was added to all wells and microplate was kept in a shaking incubator at 37° C for 48 h.

After this time, the formed biofilm was washed again with 100 μl of phosphate buffered saline. Then 100 μl of suspension of biosynthesized silver nanoparticles with concentration of 0.37×10^8 Particles ml^{-1} and 100 μl of medium were added to the wells and incubated for 24 h. The amount of the formed biofilm was measured using MTT assay. Thus, 50 μl of MTT solution was mixed with with 50 μl cell suspension and 50 μl of medium and the mixture was kept at 37 ° C for 24 h.

After washing the wells with phosphate buffered saline 100 μl of dimethyl sulfoxide was added to the wells and the wells were kept at 37° C for 15 minutes. Finally, the optical absorption of micro plates was measured with ELISA reader at a wavelength of 490 nm. At the same time, microscopic observations were done at the presence of the concentration 0.37×10^8 Particles ml^{-1} .

For this purpose, the silver nanoparticles with concentration of 0.37×10^8 Particles ml^{-1} were mixed with yeast cells and the effects of nanoparticles at 0 and 24 hours were examined using a stereo microscope. For control group the yeast cells without any nanoparticles were used [11].

Results

Spectrophotometric analysis

Figure 1 shows the ultraviolet- visible spectrum of nanoparticles synthesized using *Matricaria chamomilla*. According to this figure an absorption peak is created in the range of 200 -800 nm which indicates a surface Plasmon vibration at wavelength of 420-450 nm compared to the control group and verifies the special and quantum properties of nanoparticles synthesized by *Matricaria chamomilla*. This spectroscopy is consistent with the findings of other researches in which silver nanoparticles were synthesized using plant.

Electron microscopic study of silver nanoparticles

Figure 2. Shows electron microscopy image of silver

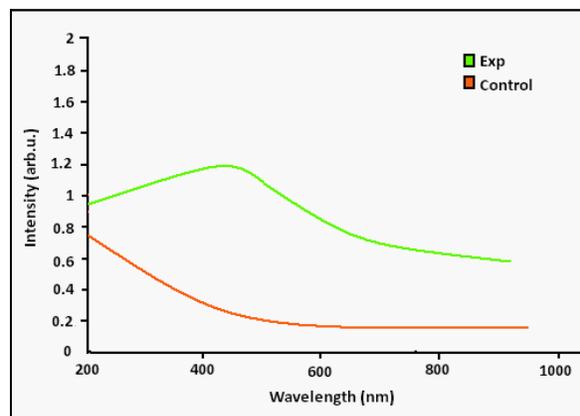


Figure 1. Ultraviolet - visible spectrum of silver nanoparticles synthesized with chamomile.

nanoparticles synthesized with *Matricaria chamomilla*. As it can be seen in the figure synthesized nanoparticles have dimensions of 60-65 nm and are in different forms which are synthesized in the form of extracellular synthesis. Since the nanoparticles are synthesized in extracellular way, reduction of costs of obtaining nanoparticles in this method a very important advantage compared with chemical synthesis of nanoparticles. Data from electron microscopy shows that the nanoparticles are produced in different sizes and this is one of the limitations of producing nanoparticles by method. However, by controlling the experimental conditions, including changes in PH and temperature and controlling the used doses of reactants, such as reducing solution (silver nitrate) and reductant (extract), this limitation can be overcome.

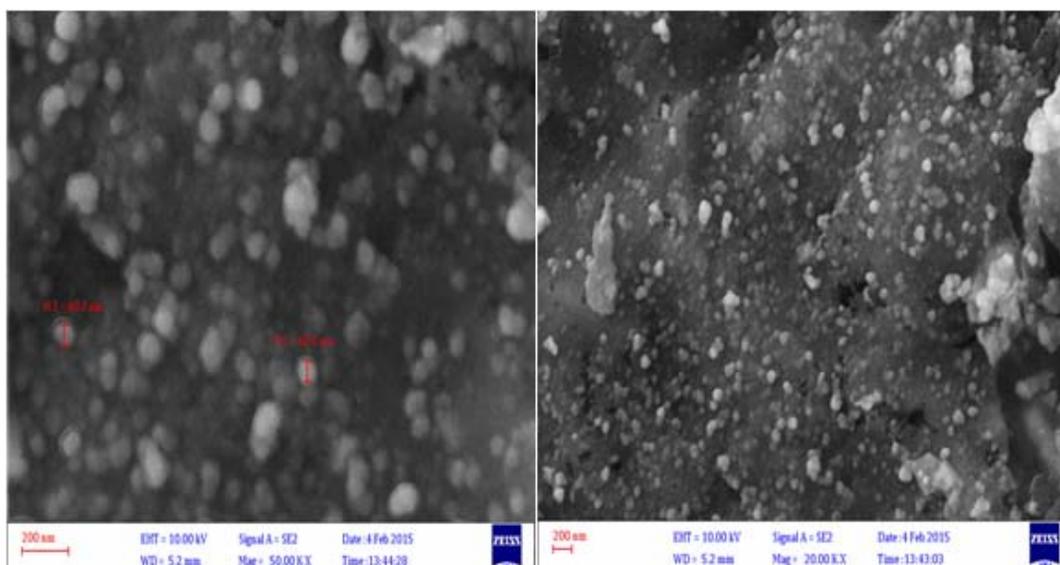


Figure 2. Electron microscopy image of biosynthesized silver nanoparticles

Exploring the antibacterial effect of the produced nanoparticles

Figure 3 shows the effects of silver nanoparticles against bacteria *Staphylococcus aureus* in broth medium. As shown in Figure 3, optical absorption of bacterial suspension containing biosynthesized silver nanoparticles during 0 to 48 hour time at the presence of nanoparticles is significantly reduced compared with the control group.

Exploring the antifungal effect of nanoparticles produced

In the study of antifungal effect of synthesized nanoparticles it was revealed that these nanoparticles have favorable effects on biofilm formation in pathogenic yeast *Candida albicans*. Figure 4 shows optical absorption of microplate at 490 nm wavelength. As it is evident in the picture, silver nanoparticles have reduced optical absorption of microplate compared with the control group. This effect is shown in Figure 5. As it is evident in the picture, nanoparticles synthesized from *Matricaria chamomilla* result in a marked reduction in the biofilm of the yeast compared with the control group. After 24 hours of treatment of yeast cells with synthesized silver nanoparticles the density and size of the cells were reduced.

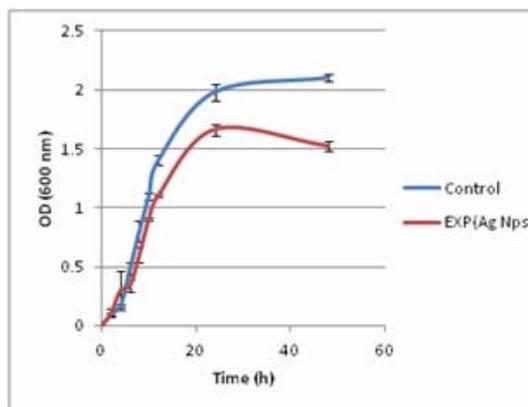


Figure 3. The antibacterial properties of biosynthesized nanoparticles against *S. aureus* ATCC (25923).

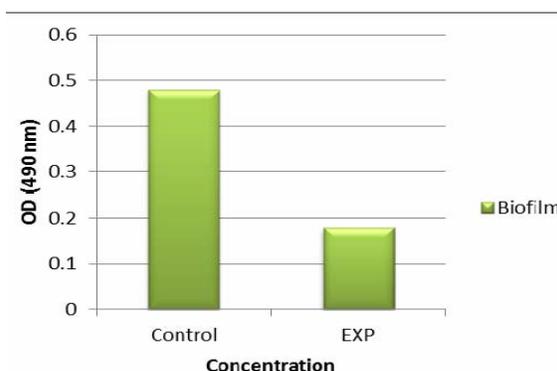


Figure 4. Optical absorption of microplate containing silver nanoparticles

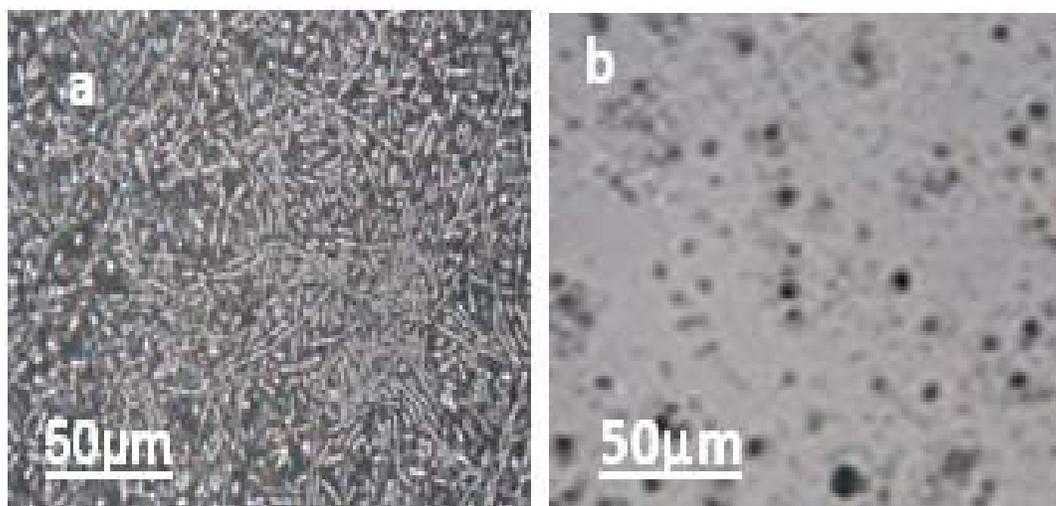


Figure 5. The microscopic observation of biofilm formation of *candida albicans* (ATCC 14053). a (control). b (treatment).

Discussion

Toxic effects of nanoparticles produced using chemical method are a serious threat to human health and a potential risk to the environment. Nanoparticles produced with biological and environment-friendly methods, can prevent the potential risks of chemical methods on humans and the environment [12]. Production of nanoparticles using live biological systems such as microorganisms (bacteria and fungi) and plants has been reported by many researchers. So far, most of the studies on the production of nanoparticles in the world have been done using bacteria and fungi [13]. In one study it was shown that *Lactobacillus* strains at exposure of silver nitrate solution result in the creation of silver nanoparticles in periplasmic space of the bacteria [14]. In one study it was concluded that *Verticillium* can result in the reduction of silver nitrate and production of silver nanoparticles with the size of 5 to 50 nm [15]. However, the production of nanoparticles by plants, especially medicinal plant is one of the methods that, in addition to the advantages over chemical methods, result in the reduction of risk and prevention of infection by microorganisms. In addition, the synthesis of nanoparticles using plants will be advantageous over other methods in terms of compatibility with the environment. In this study, production of silver nanoparticles using *Matricaria chamomilla* was evaluated for the first time. The results obtained in this study are consistent with other researches on production of silver nanoparticles by plants [16].

According to reports, UV-Vis spectra recorded for biosynthesized silver nanoparticles are in the range of 400-450 nm. In this research too, the results of UV-Vis spectra indicate the increase of surface plasmon vibrations at 420 - 450 nm wavelength. Scanning Electron Microscopy data indicate that silver nanoparticles size range is from 60 to 65 nm and with different morphologies. The results in this study are consistent with other studies [17, 18]). Shankar and *et al* synthesized silver nanoparticles from *Pelargonium* leaves. Their results showed that the plant is able to produce nanoparticles with dimensions of 16-40 nm [19]. In another study, Karimi and Mohsen Zadeh synthesized silver nanoparticles from *Achillea millefolium*. Their research showed that *Achillea millefolium* is able to synthesize silver nanoparticles with dimensions of 39-226 nm [8]. Compared to the aforementioned study, our data showed that *Matricaria chamomilla* synthesize silver nanoparticles in the size range of 60-65 nm. Silver nanoparticles have wide applications including being antibacterial and anti-fungal agents, as a catalyst in chemical reactions and biological, use in electrical batteries, use in nonlinear optics and etc. [20]. Extracellular production of nanoparticles is one of the essential and important parameters in biosynthesis of nanoparticles as in this way

achieving nanoparticles are easy and costs are lower for their purification [21]. Production of silver nanoparticles using plants results in the reduction of costs of obtaining nanoparticles. The biosynthesized nanoparticles, like chemically-synthesized nanoparticles, have the specific physicochemical properties needed for use in medical and industrial purposes and compared to with chemically-synthesized nanoparticles, biosynthesized nanoparticles have been highly paid attention to in recent years due to ease of production, repeatability, high volumes of production and the lack of production of pollution and waste. In this study it was tried to produce silver nanoparticles in extracellular form with the aim of creating optimal physicochemical conditions compared with chemical methods. *Matricaria chamomilla* is a suitable and environment-friendly reducing agent for the production of nanoparticles and this study can pave the way for industrial production of silver nanoparticles by *Matricaria chamomilla* and other plants in the future.

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Correspondence to:

Ghasem Rahimi
Young Researchers and Elite Club
Marvdasht Branch
Islamic Azad University
Marvdasht, Iran