Synthesis of 4-arylpolyhydroquinoline derivatives and evaluation of their anti-inflammatory on endometritis.

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Abstract

Four novel 4-arylpolyhydroquinoline derivatives (1-4) were synthesized and characterized via IR, 1H NMR, HRMS, and single crystal X-ray crystallography. The experimental results of anti-inflammatory activity showed that compared with compounds 1 and 2, compounds 3 and 4 with electron-repelling groups in the phenyl ring exerted rather potent activities.

Keywords: 4-Arylpolyhydroquinoline, Crystal, Anti-inflammatory.

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Introduction

Endometritis is inflammation of the endometrium, the inner lining of the uterus. Pathologists have traditionally classified endometritis as either acute or chronic [1]. Acute endometritis characterized by the presence of microabscesses is or neutrophils within the endometrial glands, while chronic endometritis is distinguished by variable numbers of plasma cells within the endometrial stroma [2,3]. The most common cause of endometritis is infection. Symptoms include lower abdominal pain, fever and abnormal vaginal bleeding or discharge. Caesarean section, prolonged rupture of membranes labor multiple and long with vaginal

Figure 1. Chemical structures of compounds 1-4.

Material and Methods

Apparatus and materials

IR spectra (400-4000 cm⁻¹) were obtained using a Brucker Equinox-55 spectrophotometer. ^{1}H NMR spectra were

examinations are important risk factors. Treatment is usually with broad-spectrum antibiotics [4].

Quinoline derivatives are valuable intermediates in the preparations of a wide range of biologically active compounds such as anticancer agents, anti-inflammatory agent, and hedgehog antagonist, etc [5,6]. In addition, they also represent a class of functionalized and versatile building blocks, that is, compounds converted these can be into 4arylpolyhydroquinoline derivatives [7,8]. With this in mind we of report here the synthesis four Λ_{-} new arylpolyhydroquinoline derivatives (Figure 1) along with their anti-inflammatory activity.



obtained using a Varian Inova-400 spectrometer (at 400 MHz). Mass spectra were obtained using a micrOTOF-Q II mass spectrometer. The melting points were taken on a XT-4 micro melting apparatus, and the thermometer was uncorrected.

Synthesis and characterization of compounds 1-4

Compounds 1-4 were synthesized according to a reported procedure [9]. A mixture of cyclohexane-1,3-dione (10 mmol), aromatic aldehydes (10 mmol), ammonium acetate (10 mmol) and ethyl acetoacetate (10 mmol) in ethanol (100mL) was refluxed for 2-3 h and then cooled to room temperature. After filtering the precipitates, they were sequentially washed with ice-cooled water and ethanol and then dried under a vacuum.

4-(3-Bromo-phenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylic acid ethyl ester (1): 235-236°C. Yield: 65.4 %. IR (KBr pellet cm⁻¹): 3265, 1650, 1600 cm⁻¹. ¹H NMR (DMSO-*d*6, δ , ppm): 9.22 (s, 1H), 7.27-7.29 (m, 2H), 7.12-7.19 (m, 2H), 4.86 (s, 1H), 3.95-4.02 (m, 2H), 2.47-2.49 (m, 2H), 2.29 (s, 3H), 2.18-2.22 (m, 2H), 1.88-1.92 (m, 1H), 1.73-1.77 (m, 1H), 1.12 (s, 3H). HRMS (ESI⁺): *m/z*: calcd for C₁₉H₂₀BrNO₃: 412.0519 [M+Na⁺]; found: 412.0522.

4-(3-Nitro-phenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylic acid ethyl ester (2): 225-226°C. Yield: 60.3 %. IR (KBr pellet cm⁻¹): 3270, 1670, 1650 cm⁻¹. ¹H NMR (DMSO-*d*₆, δ , ppm): 9.32 (s, 1H), 7.97-7.98 (d, 2H), 7.60-7.61 (d, 1H), 7.51-7.54 (t, 1H), 5.00 (s, 1H), 3.96-3.98 (t, 2H), 2.50-2.52 (m, 2H), 2.32 (s, 3H), 2.16-2.26 (m, 2H), 1.89-1.93 (m, 1H), 1.74-1.77 (m, 1H), 1.10-1.12 (t, 3H). HRMS (ESI⁺): *m/z*: calcd for C₁₉H₂₀N₂O₅: 379.1266 [M+Na⁺]; found: 379.1879.

4-(3-Methoxy-4-hydroxy-phenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydro-quinoline-3-carboxylic acid ethyl ester (3): 230-231°C. Yield: 58.4 %. IR (KBr pellet cm⁻¹): 3260, 1669, 1632 cm⁻¹. ¹H NMR (DMSO- d_6 , δ , ppm): 9.07 (s, 1H), 8.62 (s, 1H), 6.70-6.71 (d, 1H), 6.57-6.58 (d, 1H), 6.47-6.49 (q, 1H), 4.80 (s, 1H), 3.99-4.01 (q, 2H), 3.68 (s, 3H), 2.46-2.48 (m, 2H), 2.25 (s, 3H), 2.18-2.21 (m, 2H), 1.89-1.92 (m, 1H), 1.75-1.78 (m, 1H), 1.14-1.17 (t, 3H). HRMS (ESI⁺): *m/z*: calcd for C₂₀H₂₃NO₅: 380.1477 [M+Na⁺]; found: 380.1460.

4-(4-Methoxy-phenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylic acid ethyl ester (2): 228-229°C. Yield: 68.2 %. IR (KBr pellet cm⁻¹): 3212, 1678, 1600 cm⁻¹. ¹H NMR (DMSO-*d*₆, δ , ppm): 9.09 (s, 1H), 7.03-7.04 (d, 2H), 6.73-6.74 (d, 2H), 4.82 (s, 1H), 3.97-3.98 (d, 2H), 3.67 (s, 3H), 2.45-2.48 (q, 2H), 2.27 (s, 3H), 2.14-2.20 (m, 2H), 1.87-1.91 (m, 1H), 1.73-1.75 (m, 1H), 1.12-1.14 (t, 3H). HRMS (ESI⁺): *m/z*: calcd for C₂₀H₂₃NO₄: 364.1510 [M+Na⁺]; found: 364.1577.

Crystal structure determination

Suitable single crystals of compound 4 were obtained by evaporation of chloroform solution. The diffraction data were collected on a Bruker Smart Apex CCD area detector using a graphite monochromated Mo K α radiation (λ =0.71073 Å) at room temperature. The structure was solved by using the program SHELXL-97 [10] and Fourier difference techniques, and refined by full-matrix least-squares method on F^2 . All

hydrogen atoms were added theoretically. Crystallographic data for compound 4 are listed in Table 1.

Table 1. Crystal data, data collection and structure refinement of compound 4.

Formula	C ₂₀ H ₂₃ NO ₄
Mr	341.39
Temperature/K	293 (2)
Crystal system	Triclinic
Space group	Pī
a/Å	7.2610 (19)
b/Å	10.511 (3)
c/Å	12.620 (3)
a/°	74.305 (4)
β/°	73.718 (4)
γ/°	82.747 (5)
V/Å ³	888.7 (4)
Z	2
D _{calc} /g⋅cm ⁻³	1.276
μ(Mo Kα)/mm ⁻¹	0.089
θ range/°	1.73 to 25.00
Reflections collected	4575
No. unique data [R(int)]	3115 [0.0215]
No. data with $l \ge 2\sigma(l)$	2267
<i>R</i> ₁	0.0574
$\omega R_2(all data)$	0.1762
CCDC	1469526

Anti-inflammatory activity

Anti-inflammatory activity was measured by carrageenan induced paw edema as described by Winter et al. Drugs were given orally as solution is distilled water. The solution was so prepared that for every 100 g body weight of the animal 0.3 to 0.8 mL of the solution was administered depending on the dose. Groups of six rats were dosed orally with drug 1h before injection of 0.1 mL of a 1 % suspension of carrageenan in normal saline into the subplantar region of the right hind paw. Control animals (six) received carrageenan only. Oedema was measured 3h later plethysmographically. Mean increase in the paw volume and Standard Error of the Mean (SEM) for each group were calculated and the results were expressed as percent inhibition of oedema as compared to the control group. ED_{50} values were determined by semilogarithmic plot of the % inhibition versus log dose.

Results and Discussion

Molecular structure

The ¹H NMR, IR and MS for the products are in good agreement with the title compounds. In the ¹H NMR spectra of compounds 1-4, single peaks at δ 4.86, δ 5.00, δ 4.80 and δ 4.82 ppm characteristic of the CH group respectively. In order to confirm the configuration of the product, single crystals of compound 4 were cultured for X-ray diffraction analysis. The crystal structure of compound 4 was presented in figure 2. The six-membered ring containing nitrogen atom is nearly planar [maximum deviation=0.1734 Å] and the adjacent ketone ring adopts a flattened chair conformation [dihedral angle (C₅-C₆-C₇-C₈)=16.303°]. The nitrogenous heterocyclic is almost perpendicular to the benzene ring [dihedral angle=85.435°] and is almost coplanar with the mean plane of the ketone ring [dihedral angle=11.343°].



Figure 2. Molecular structure of compound 4.

	Antiinflammatory activity (%) ± SME			
Compounds	Dose (mg/kg)		ED ₅₀ (mg/kg)	
	25	50	100	
1	0.7 ± 1.8	21.8 ± 1.3 [*]	29.0 ± 2.1	189.5
2	0.9 ± 1.9 [*]	20.2 ± 1.7	26.3 ± 2.6	193.5
3	20.1 ± 2.3	55.3 ± 2.1*	66.2 ± 3.7	61.2
4	21.2 ± 2.1*	59.9 ± 2.7	61.5 ± 3.5 [*]	57.2

Anti-inflammatory activity by carrageenan induced paw edema in rats at the end of 3h. Drug was given orally as solution in distilled water (*p<0.05 Mann-Whitney test). Six rats were used for each test group and the control.

Biological activity

The anti-inflammatory activity of compounds 1-4 were evaluated by carrageenan induced paw edema in rats, and the ED_{50} values derived from the experimental data were summarized in table 2. As can be seen in table 2, there is great

difference in the anti-inflammatory activity among the four compounds. Compared with compounds 1 and 2, compounds 3 and 4 showed more potent anti-inflammatory activity with ED_{50} value of 61.2 and 57.2 mg/kg, which is much lower than those of compounds 1 and 2.

Conclusion

In conclusion, we synthesized four novel 4arylpolyhydroquinoline derivatives and characterized them via IR, ¹H NMR, HRMS, and single crystal X-ray crystallography. We have found that compared with compounds 1 and 2, compounds 3 and 4 exhibited the better anti-inflammatory efficiency with ED_{50} values of 61.2 and 57.2 mg/kg, possibly because of the electron-repelling group in the phenyl ring in the structures of compounds 3 and 4.

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