

## Stability of bovine serum albumin labelled by rhodamine B isothiocyanate.

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### Abstract

The aim of this study was to label Bovine Serum Albumin (BSA) by Rhodamine B Isothiocyanate (RBITC) and test its stability *in vivo*. RBITC-BSA was labelled by the improved Marshall's method, scanned at full wavelength, and the RBITC:BSA ratio was calculated. Blood and urine samples were obtained from rabbits injected with the RBITC-BSA complex and SDS-PAGE analysis was performed. Full wavelength scanning showed that the labelled RBITC-BSA complex possessed features of both RBITC and BSA, and the RBITC:BSA ratio was 1:80. Furthermore, the RBITC-BSA complex was stable *in vivo*. Thus, we developed a stable RBITC-BSA complex with high specificity and sensitivity, which could be used as a tracer molecule to study protein transportation and vascular permeability.

**Keywords:** Rhodamine B isothiocyanate, Bovine serum albumin, Full wavelength scanner, SDS-PAGE.

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### Introduction

With the recent development of fluorescence-based analytical approaches in medicine and biochemistry, several fluorescent probes are being used in biological analysis. Presently, fluorescent probes include small molecules [1-4], nanoprobe [5-7], luminescent lanthanide complexes [8-10], and certain fluorescent proteins [11-14]. Protein-based fluorescent probes are used to detect physiological activities of proteins, which is important for clinical treatment. Since the first report of a fluorescent protein antibody in 1950 by Coons and Kaplan [15], there has been an increase in the development and use of fluorescent small molecule-protein combinations in bioanalysis, which has played pivotal roles in biomedical and basic research.

Fluorescent isothiocyanate is a common fluorescent probe, which has high quantum yield, superior light stability, and low temperature coefficient. Under alkaline conditions, the isothiocyano group of the fluorescent isothiocyanate could directly combine with the amino moiety of proteins via a phosphoryl reaction to form a fluorescein-protein complex. Fluorescein-albumin is as fluorescent tracer molecule, which is used to investigate vascular permeability [16]. A fluorescein-albumin complex developed by a statistically significant method must possess the following characteristics: 1) the fluorescein must be conjugated to the protein *in vitro*; 2) the fluorescein-albumin complex should be separable from the non-reacted fluorescein and its degradation products [17].

However, studies reporting the stability of the RBITC-BSA complex as a fluorescent tracer *in vivo* are lacking. To investigate whether the RBITC-BSA complex was suitable as a tracer molecular *in vivo*, we designed an RBITC-BSA complex; the RBITC was removed by glucose gel chromatography and analysed by full wavelength scanning. Moreover, serum and urine samples from rabbits at different time points were analysed to determine the stability of the RBITC-BSA by Sodium Dodecyl-Polyacrylamide Gel Electrophoresis (SDS-PAGE).

### Materials and Methods

#### Preparation of the RBITC-BSA complex

BSA was labelled by the improved Marshall's method. Saline (0.15 M, NaCl) and a buffer solution (0.15 M, NaHCO<sub>3</sub>/Na<sub>2</sub>CO<sub>3</sub>, pH 9.0) were mixed in a 9:1 (v/v) ratio to form a 10 mg/ml protein solution. Then, RBITC, in which the M ratio of protein to fluorescein was 5:1, was added slowly into the protein solution, followed by stirring for 12 h in the dark at 4°C. After that, the mixed solution was placed in a filter bag and dialyzed against phosphate buffer saline (0.01 M PBS, pH 7.4) at 4°C in the dark for 2 days.

A Sephadex G-75 column (Superfine, GE Healthcare, 2.6 cm diameter, 100 cm length) was equilibrated with phosphate buffer (8 g/L NaCl, 0.2 g/L KCl, 0.2 g/L KH<sub>2</sub>PO<sub>4</sub>•12H<sub>2</sub>O, pH 7.4), followed by loading of specimens at 4°C, and dialysis in the dark. The column was washed with PBS at a speed of 10

min/3 ml (constant flow pump BT100-1F, Baoding Lange Constant Flow Pump Co., Ltd) and 3 ml eluates were collected per tube (automatic collector BS-100A, Shanghai Qingpuhuxi instrument factory). The eluate corresponding to the first fluorescent band from the column was observed and recorded. After collecting approximately 500 ml of eluate, each tube was examined by an Ultraviolet (UV) spectrophotometer (Ultrospec 2100 pro, GE) and absorbance values at 280 nm were calculated.

### Confirmation of the light absorption characteristics

In the RBITC-BSA complex, the RBITC and BSA were subjected to full wavelength scanning (190-900 nm) by the UV-visible spectrophotometer. The changes in the light absorption characteristics of RBITC-BSA, RBITC, and BSA were compared.

### Detection of labelling rate

The protein ( $\lambda=230$  nm) and fluorescein levels (Ex: 555 nm, Em: 582 nm) in the eluting peak were detected by the UV-visible spectrophotometer and quantitated by the standard curve method. The formula to calculate the molecular ratio of RBITC and BSA in the product was  $F/P = (\text{fluorescein level } (\mu\text{g/ml}) \times 10^{-3} / \text{fluorescein molecular weight}) / (\text{protein level (mg/ml)} / \text{albumin molecular weight})$ .

### Preparation of specimens in vivo

Three milliliters of the RBITC-BSA complex was injected in healthy male Japanese rabbits weighing 2.5 kg. Thereafter, 2 ml blood from the right ear vein (experimental) of the rabbits was collected at 10 min and 30 min after the injection, respectively. Similar amount of blood was drawn from the left ear vein as a control before the injection. Blood specimens were marked as 0, 10, and 30 that corresponded to 0 min, 10 min, and 30 min samples, respectively. Two milliliters urine was obtained from bladder.

### SDS-PAGE analysis

Five microliters specimen (about 30-50  $\mu\text{g}$  protein) was used for SDS-PAGE on a 12% gel at 120 V for 80 min. First, the fluorescent signal on the gel was imaged, followed by staining with Coomassie Brilliant Blue R250 for 1 h. After that, the gel was destained overnight. Finally, the gel was photographed by an electrophoretic image analyzer (Bio-RAD Universal HoodII)

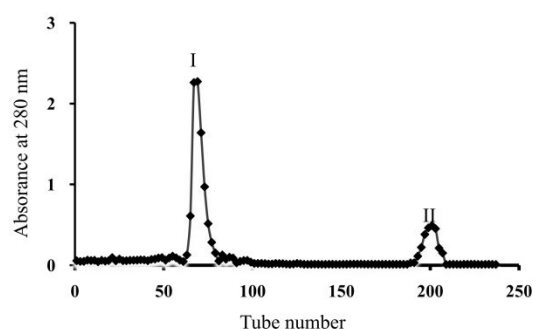
## Results

### Chromatography

The 60<sup>th</sup> eluate tube of the Sephadex G-75 column chromatography of the RBITC-BSA complex corresponded to the appearance of the first fluorescent band. There were two eluting peaks corresponding to the Optical Density 280 (OD280) values (Figure 1); the first peak occurred in fractions

60-80 (eluting peak I) and the second peak in fractions 180-220 (eluting peak II). Based on the principle of glucose gel chromatography, the eluting peak I corresponded to the RBITC-labelled BSA, which was collected. The eluting peak II was free RBITC, which was discarded.

The OD280 readings of the eluate fractions showed two values in the UV-visible spectrophotometer; the first value corresponded to tube numbers 60-80 (eluting peak I), which contained the RBITC-labelled BSA; these fractions were combined; the second value corresponded to tube numbers 180-220 (eluting peak II), which was free RBITC, and were discarded.



**Figure 1.** Sephadex G-75 glucose gel chromatography of RBITC-BSA complex.

### Verification of the light absorption characteristics of RBITC

The maximal (at 555 nm) and minimal photoabsorption (at 197 nm) of RBITC (Figure 2A) and the maximum photoabsorption of BSA at 555 nm (Figure 2B) were assessed by full wavelength scanning. However, the eluting peak I had two maxima of photoabsorption at 230 nm and 558 nm, respectively (Figure 2C), which indicated that the eluting peak was the RBITC-labelled BSA complex.

RBITC had the maximal light absorption at 555 nm and minimal light absorption at 197 nm (Figure 2A), whereas BSA had the maximal light absorption at 230 nm (Figure 2B). The fractions containing the eluting peak I (Figure 1) had two maxima of light absorption at 230 nm and 558 nm, thereby confirming the presence of the RBITC-BSA complex (Figure 2C).

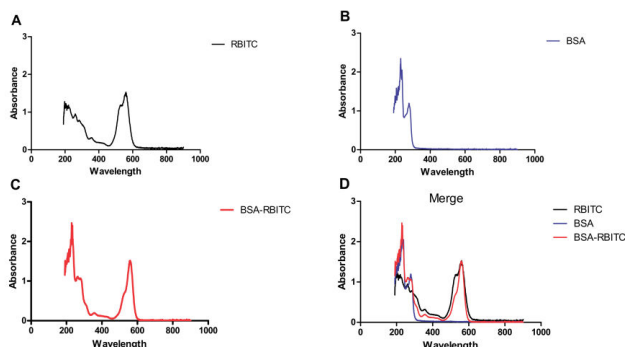
### Calculation of the marked rate

The F/P value of fractions 60-80 (first eluting peak) of the Sephadex G-75 gel chromatography was 1.80. The F/P value reflects the amount of protein marked by fluorescein. Generally, the value of F/P ranges from 1 to 2. A high value indicates non-specific marking, whereas low value indicates weak fluorescence and low sensitivity.

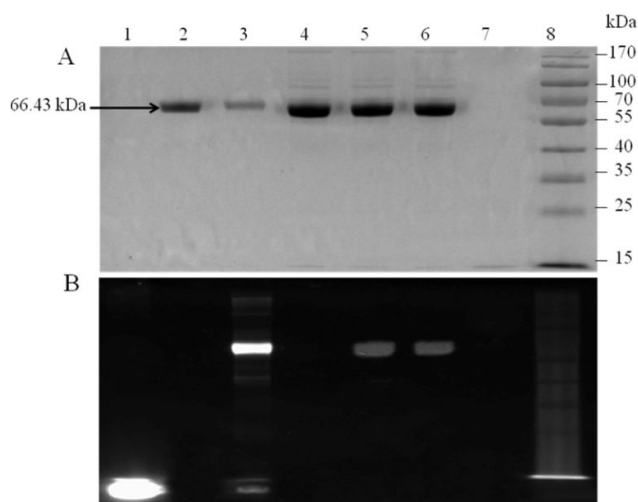
### SDS-PAGE

SDS-PAGE analysis of the serum obtained from rabbits injected with the RBITC-BSA complex showed fluorescent

signal in the 10 min and 30 min post-injection samples; the position of fluorescent band was identical to the position of the chromatographically purified RBITC-BSA (Figure 3B). This result suggested that the RBITC-BSA complex was stable before and after the *in vivo* experiment. Coomassie Brilliant Blue R250 staining showed that the purified RBITC-BSA complex and the fluorescent bands observed in the 10 min and 30 min post-injection serum samples electrophoresed at 66.43 kDa, which is the size of BSA. This indicated that BSA was also stable in the serum; however, the protein was not observed in the urine samples under identical conditions (Figure 3A).



**Figure 2.** Comparisons of photoabsorption characteristics among RBITC, BSA and RBITC-BSA complex at 190-900 nm wavelength. A: RBITC, B: BSA, C: RBITC-BSA.



**Figure 3.** Coomassie brilliant blue staining image (A) and fluorescent scanning image (B) by 12% SDS-PAGE analysis. Lane 1: RBITC; lane 2: BSA; lane 3: RBITC-BSA; lane 4: serum at 0 min; lane 5: serum at 10 min after injected RBITC-BSA; lane 6: serum at 30 min after injected RBITC-BSA; lane 7: urine, lane 8: marker.

## Discussion

Fluorescent isothiocyanate is a probe with superior fluorescence characteristics. The isothiocyanate group of fluorescent isothiocyanate can directly conjugate with the amino group of proteins to form sulfur-hydrocarbon bonds via a phosphoryl reaction to yield a fluorescently labelled protein probe. The isothiocyanate group of RBITC can bind to the amino group of proteins (mainly  $\epsilon$ -NH of lysine) under alkaline

conditions, and yield a fluorescent protein probe via the formation of a carbon-sulfur-amino bond, which involves acylation of carbon [18]. Although the tertiary structure of proteins changed at pH 9.0 during the formation of the RBITC-BSA complex, the complex still retained its secondary helical structure [19]. Andersson et al. [20] found three binding sites for fluorescein in BSA, which utilize both stable covalent and unstable non-covalent bonds. BSA could combine with RBITC via stable covalent bonds after the scrambling of its three-dimensional structure in the alkaline pH of the reaction. The non-covalent bonding between RBITC and BSA was disrupted by chloride ions of the buffer solution used in glucose gel chromatography. Full wavelength scanning indicated that the first peak obtained from the Sephadex column contained the BSA-RBITC complex, which was formed via covalent bonding. This complex was injected into rabbits; SDS-PAGE analysis of the post-injection serum samples revealed the presence of fluorescently labelled proteins, which were of the same size as that of BSA. This indicated that the RBITC-BSA complex was stable *in vivo* and could be used as a tracer in studies regarding capillary permeability and molecule transfer.

## Conclusion

We successfully prepared a RBITC-BSA complex, which is highly specific, sensitive, and stable *in vitro*, and could be used as a tracer molecule for study of protein transportation and vascular permeability.

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## Conflicts of Interest

All of the authors declare that they have no conflicts of interest regarding this paper.

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